

Automated Generation of a Dihydropyrimidine Library **Using Sequential Microwave-Assisted Synthesis**



Alexander Stadler and C. Oliver Kappe*

Institute of Chemistry, Organic and Bioorganic Chemistry, Karl-Franzens-University Graz, Heinrichstrasse 28, A-8010 Graz, Austria website: http://www-ang.uni-graz.at/~kappeco email: alexander.stadler@uni-graz.at



Optimization of Reaction Conditions using Microwaves

Model Reaction:



Conventional Conditions: EtOH, cat. HCl, reflux, 3h, 78% yield K. Folkers et al., J. Am. Chem. Soc., 1932, 54, 3751-

3758

- Step 1: Choose Solvent
 - best solvent: AcOH/EtOH 3:1
 - · effectively couples with microwaves
 - dissolves building blocks under reaction conditions
 - DHPM products sparingly solubleat rt

Step 2: Select Catalyst



 HCI causes decomposition of urea and leads to unwanted byproducts at higher temperatures · Lewis acids are more tolerable and have been reported to be effective catalysts

Step 3: Optimize Temperature & Time





⇒ General MW Protocol: 4 mmol building blocks, 2 ml AcOH/EtOH 3:1 10mol% Yb(OTf)₃, 10 min, 120°C

Step 4: Reoptimization for Troublesome Building Block Combinations





EtOH 10 mol% LaCl₃ 10 mol% Yb(OTf)₃ 120°C, 10 min 41% yield

100°C, 20 min

50% vield

EtOH 10 mol% HCl 120°C, 15 min 59% yield

Results and Conclusions

Software Aided Library Generation

Smith Workflow Manager



- Prepare stock solutions of aldehydes (AcOH) and CH-acidic carbonyls (EtOH)
- Enter building blocks & reaction conditions into the software
- Generate dispensing strategy
- Run the automated protocol (unattended)
- Work up (filter products directly or add H₂O)

- > 48 member DHPM library generated within 12h (52% average yield)
- > DHPMs produced in 200-1000 mg quantities
- Reaction times reduced from hours to minutes
- Reaction optimization within hours
- > Establishing of library production protocol within days
- > Sequential treatment allows for individuality optimized conditions

Stadler A., Kappe C.O., J. Comb. Chem., 2001, 3, 624-

630

Acknowledgements

This work was supported by the Austrian Science Fund.

We thank PersonalChemistry AB(Uppsala, Sweden) for the use of their instument.