

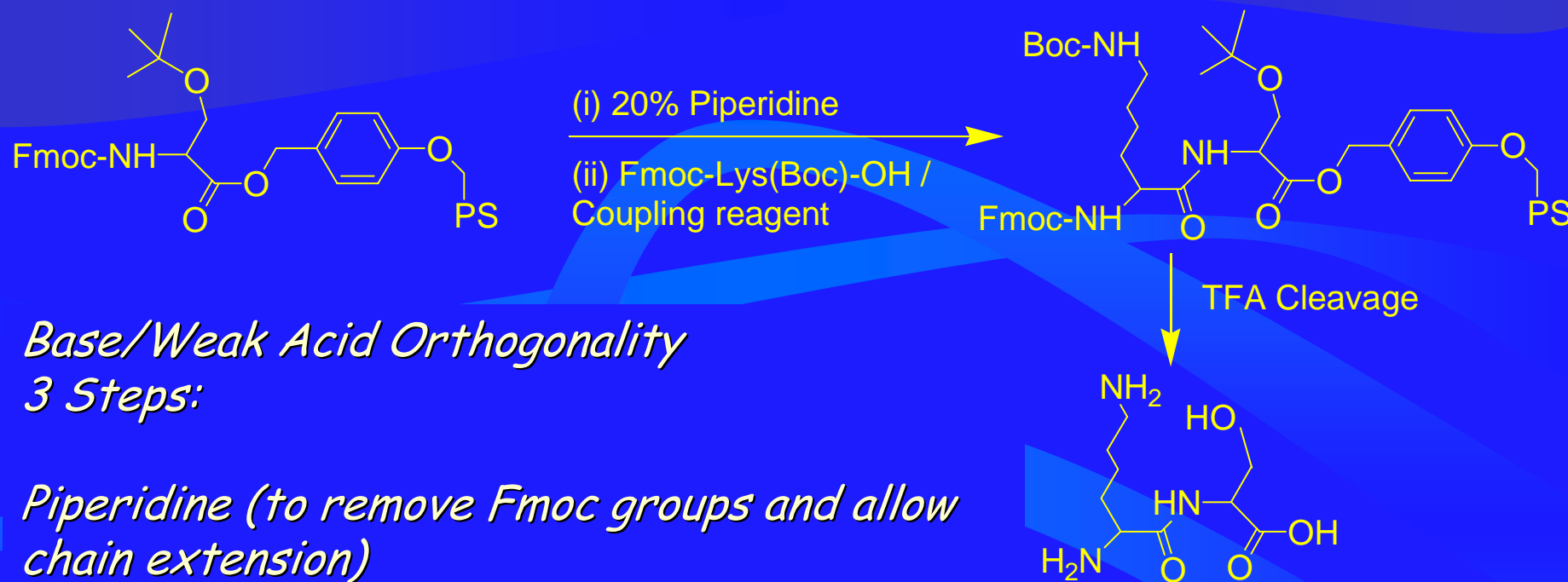
Microwave Chemistry and Solid Phase Synthesis

(a). Solid phase peptide synthesis

(b). Immobilized reagents

(c). Pd chemistry

Solid Phase Peptide Synthesis - The Fmoc/Bu^t Route



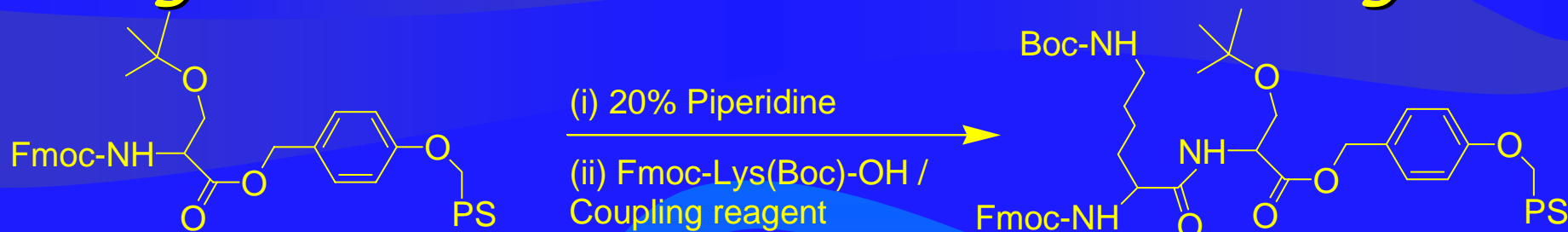
*Base/Weak Acid Orthogonality
3 Steps:*

*Piperidine (to remove Fmoc groups and allow
chain extension)*

*TFA for cleavage from the resin and side
chain deprotection!*

*Coupling of new amino acid to growing peptide
chain*

Solid Phase Peptide Synthesis - Which Steps Might Benefit from Microwave Heating?

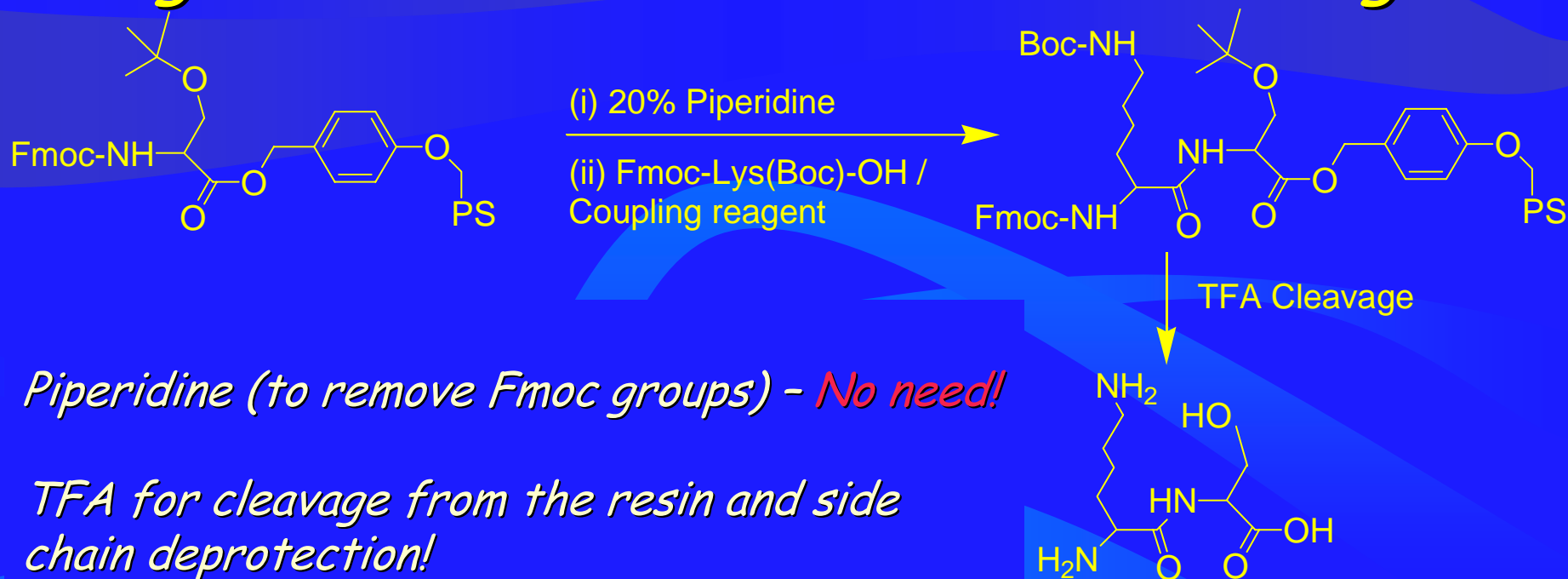


Piperidine (to remove Fmoc groups)

TFA for cleavage from the resin and side chain deprotection!

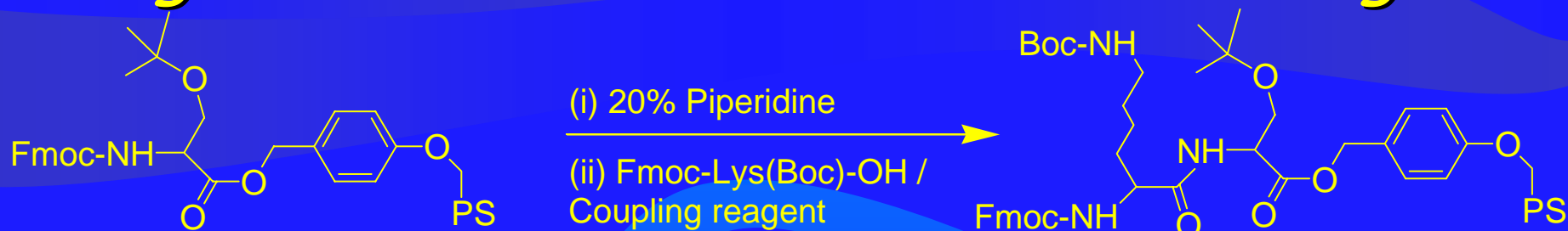
Coupling of new amino acid to growing peptide chain

Solid Phase Peptide Synthesis - Which Steps Might Benefit from Microwave Heating?

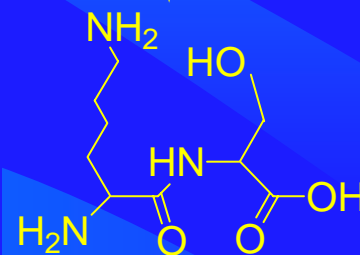


Coupling of new amino acid to growing peptide chain

Solid Phase Peptide Synthesis - Which Steps Might Benefit from Microwave Heating?



TFA Cleavage

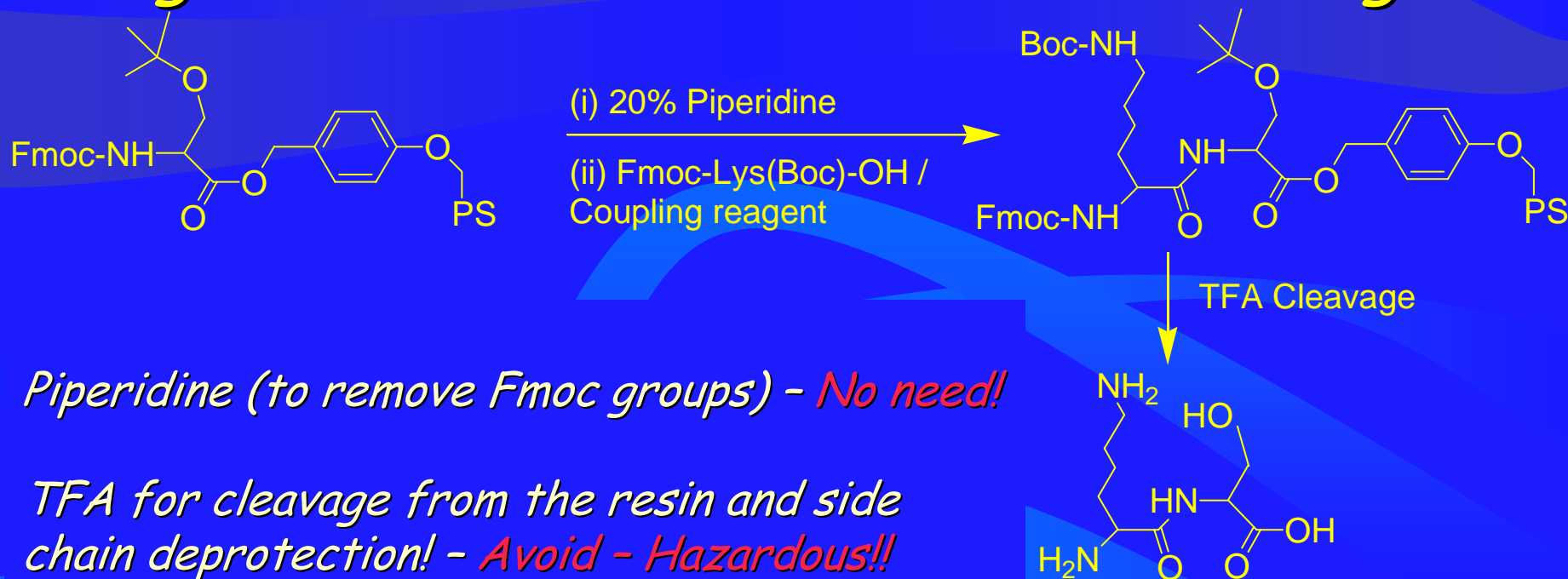


Piperidine (to remove Fmoc groups) - No need!

TFA for cleavage from the resin and side chain deprotection! - Avoid - Hazardous!!

Coupling of new amino acid to growing peptide chain

Solid Phase Peptide Synthesis - Which Steps Might Benefit from Microwave Heating?



Piperidine (to remove Fmoc groups) - *No need!*

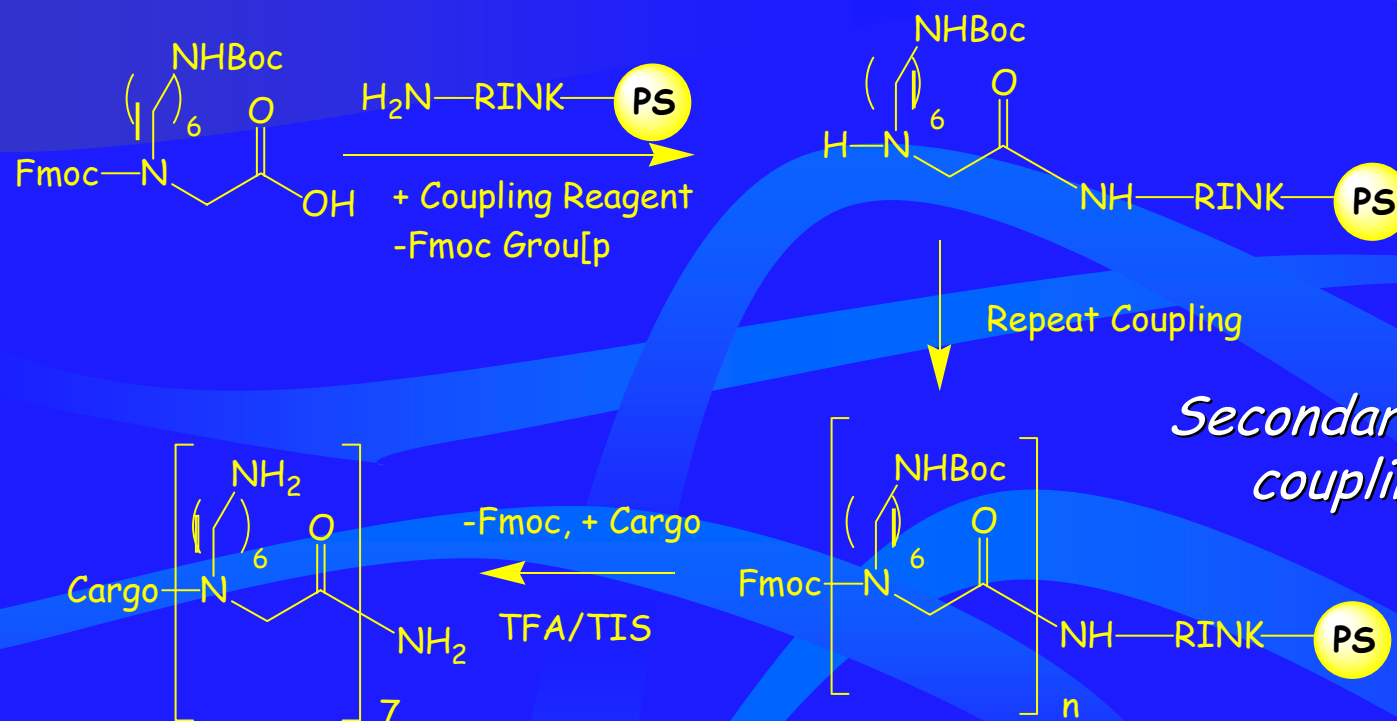
TFA for cleavage from the resin and side chain deprotection! - *Avoid - Hazardous!!*

Coupling of new amino acid to growing peptide chain - *Yes - Main cause of peptide synthesis failure*

(a). Peptide Chemistry and Microwave Heating

- Fmoc group stability and DMF decomposition*
- Many coupling agents have short half-lives once added to a reaction mixture at 25°C (see Albericio, JOC, 1998 e.g. HATU and PyBOP - $t_{1/2}$ = 40 min at 25°C - so at 65°C, $t_{1/2}$ = 2.5 min).*
- Avoid the use of exotic (and expensive) coupling reagents - side reactions, short lived*
- Do not heat Fmoc groups above 65°C*
 - DMF, DMA or NMP as solvent.*
 - Use DIC/HOBt for coupling.*

(a). Peptide Chemistry and Microwave Heating



Repeat Coupling

Secondary amine makes coupling difficult

N-Alkylglycines for cell delivery

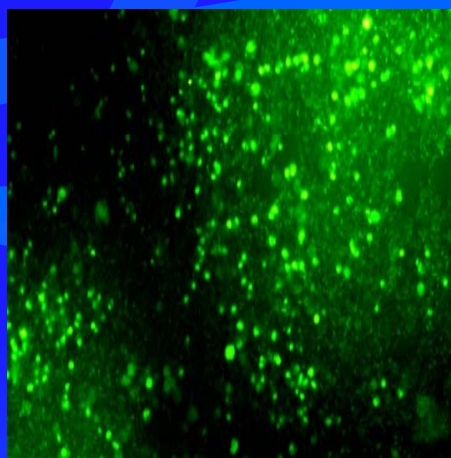
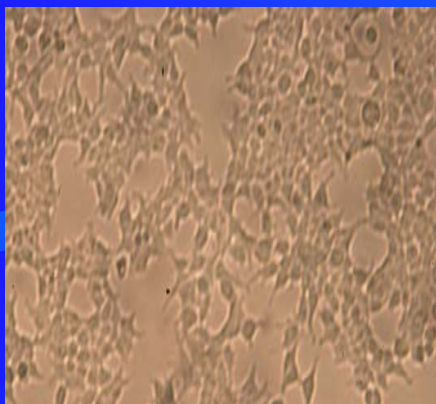
Fmoc-NH(CH2)6-COOH
 \longrightarrow
 $\left[\text{Fmoc-NH(CH}_2\text{)}_6\text{-CO-NH-RINK-PS} \right]_7$

ADC1 A, ADC1 (050905090229.D)
PMP1, Solvent B

Chromatogram showing a single sharp peak at 3.197 minutes. The y-axis is labeled 'Norm.' and ranges from 0 to 17500. The x-axis is labeled 'Time (min)' and ranges from 0 to 15. A linear ramp is overlaid on the baseline, starting at 0 and reaching 17500 at approximately 10 minutes.



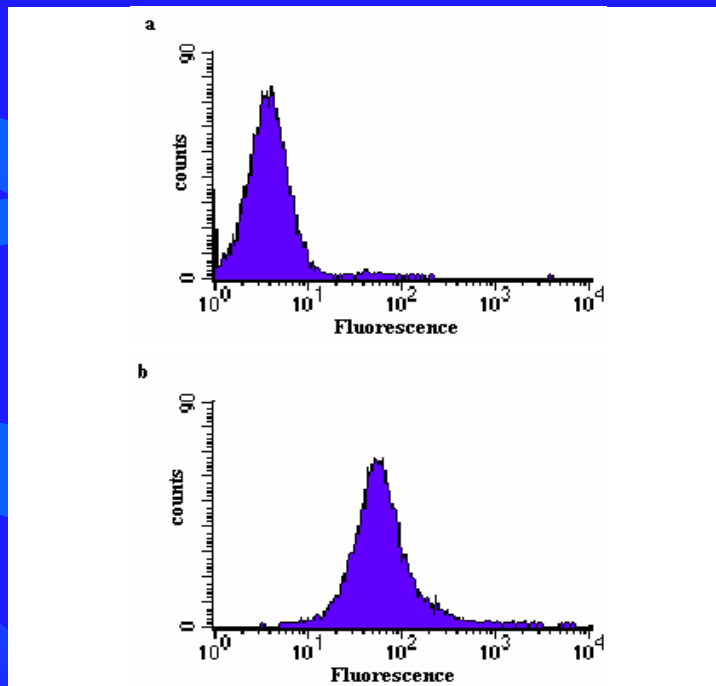
Cellular Delivery - Cell Permeable Peptoids



HEK293T (human embryonic kidney) cells, general transmission image at low magnification

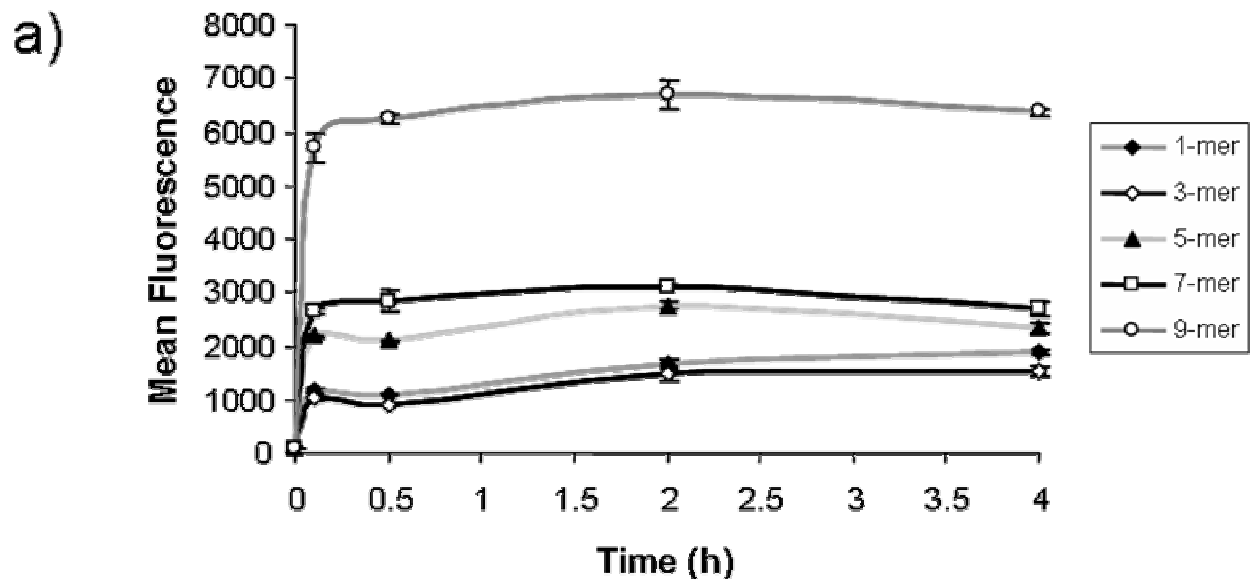
10 μM , 37°C for 6h, fluorescent image at low magnification

Flow cytometry analysis with HEK293T cells

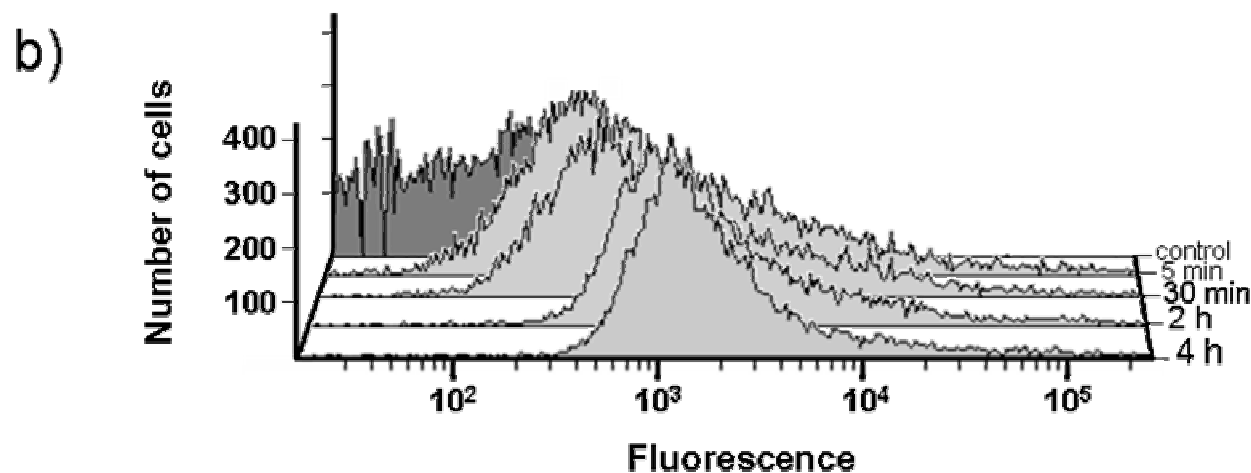


Oligomer	% Incorporation
n = 3	99
n = 5	93 (25°C)
n = 5	97 (37°C)
n = 7	99

Cell Permeable Peptoids - Uptake

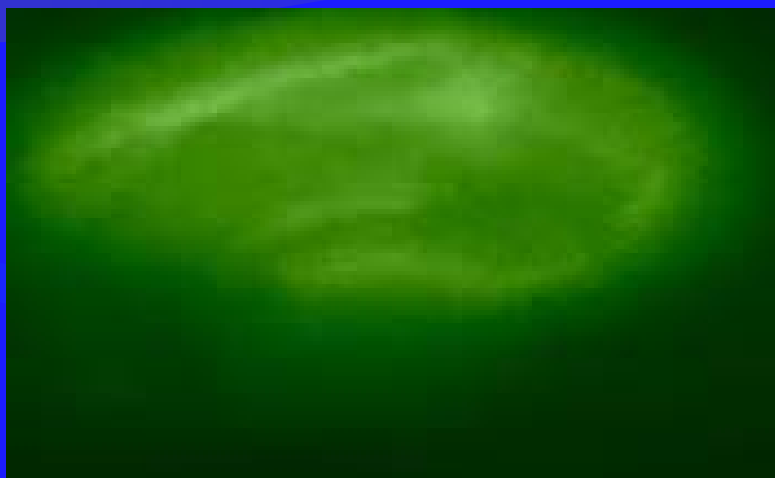


a) Comparative uptake of K562 cells with oligomers ($n=1,3,5,7,9$)

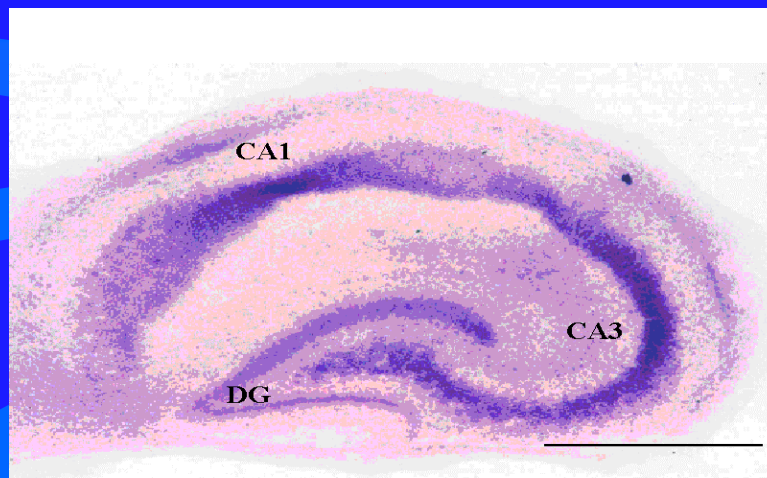


b) HeLa cells after incubation with $10 \mu\text{M}$ solution of compound ($n=9$).

Cell Permeable Peptoids



Cellular uptake of
peptoid ($n = 7$) by a
hippocampal slice
culture



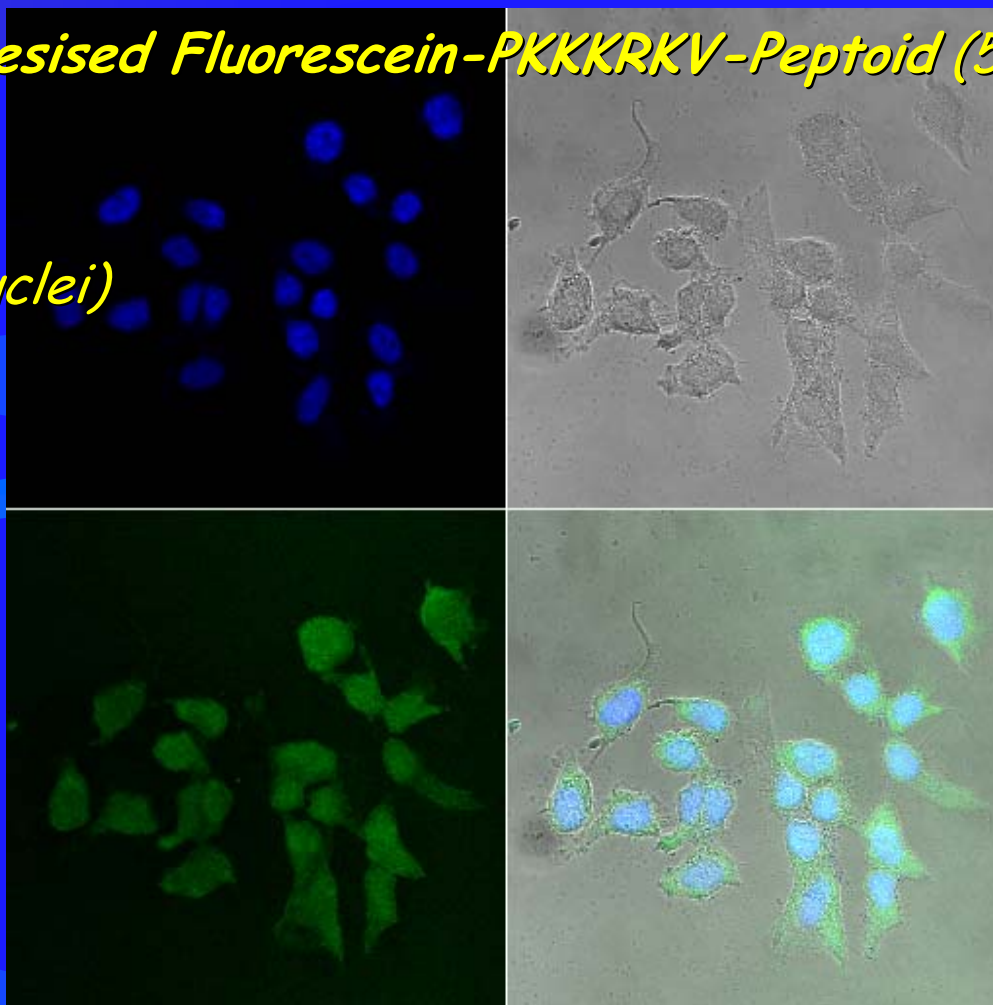
Thionin stained
hippocampal slice
showing all the cell
layers

Directed Chemistry in Cells

Nuclei labelling in live cells

Synthesised Fluorescein-PKKKRKV-Peptoid (5 mer)

(DAPI stained nuclei)



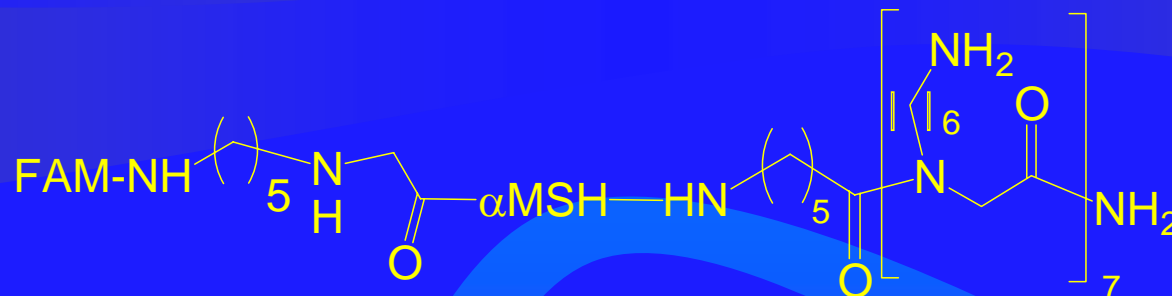
Delivery within 30 mins

Can we deliver α -MSH into Cells and Tissue?

*α -MSH : α -Melanocyte Stimulating Hormone
(Ac-SYSMEHFRWGKPV-NH₂)*

- Cell impermeable.*
- Binds the Melanocortin 1 receptor (MC1R).*
- Stimulates the pigmentation of melanocytes and melanoma cells.*
- Immunomodulatory activity (suppression of lymphocyte proliferation in vitro at 10^{-13} M to 10^{-11} M).*

(a). Peptide Chemistry and Microwave Heating



Addition of 17-mer peptide (α MSH)

*Microwave heating: 60 °C, 10 minute couplings
DIC/HOBt (0.1M, 3equiv.)*

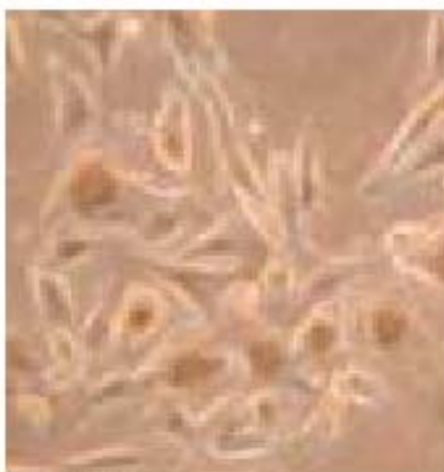
*Coupling with 5(6)-carboxyfluorescein (again 10
minutes and quantitative)*

Isolated yield of peptide = 45% (>98% per step)

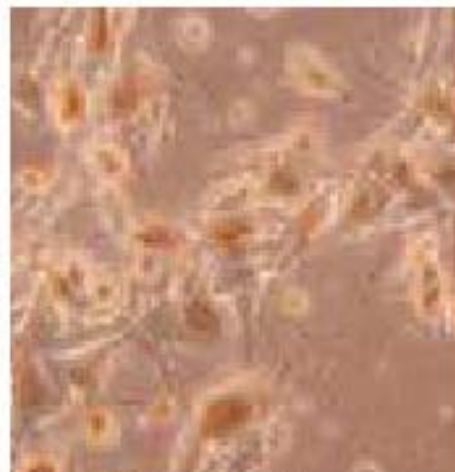
Pigmentation Assay



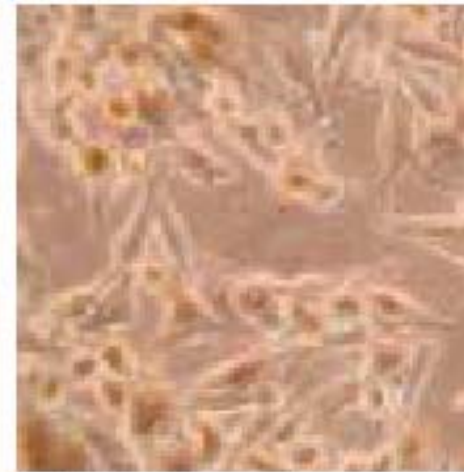
Control



α MSH



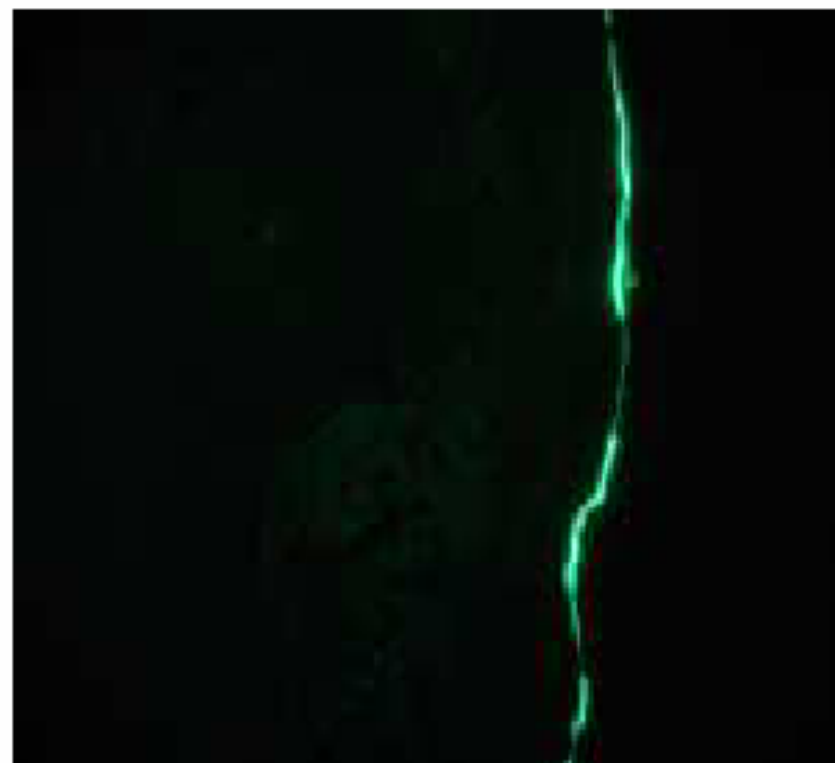
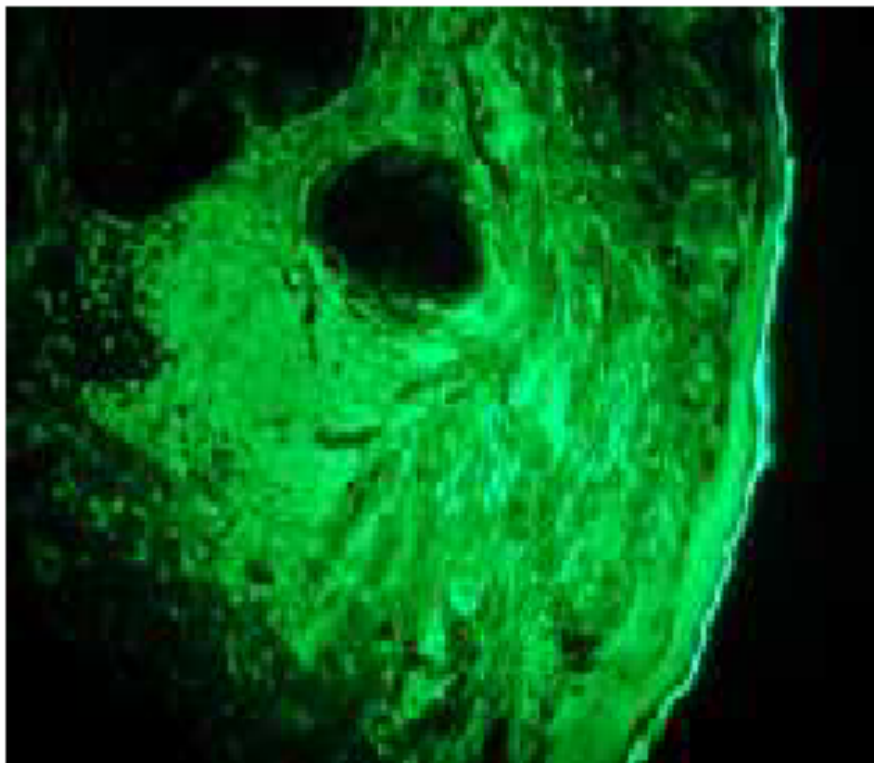
α MSH-PPL-98



α MSH-PPL-99

S91 melanoma cells in culture

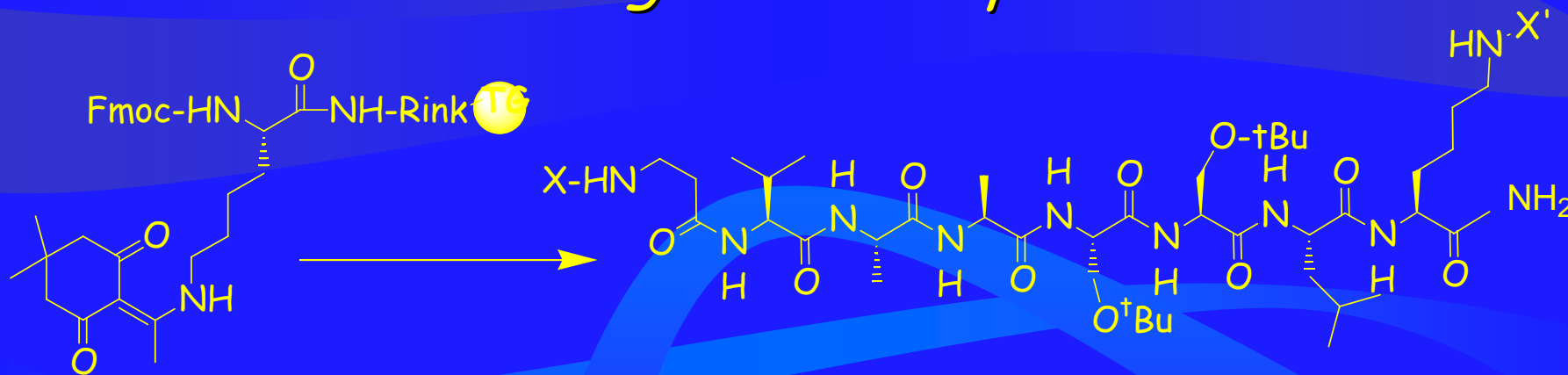
Human Skin - Delivery



Fluo-α-MSH-Peptoid (99) (3h, 10^{-3} M)

Fluo-α-MSH (3h, 10^{-3} M)

(a). Peptide Chemistry and Microwave Heating - FRET Peptides

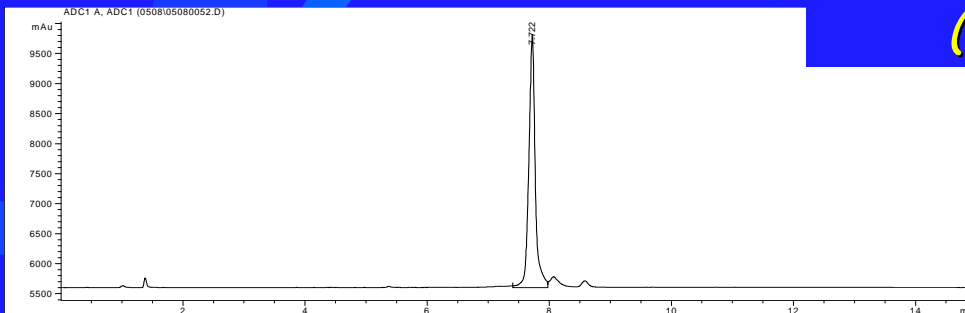


$X = \text{Dabcyl}; X' = \text{FAM}$

$X = \text{Fmoc-3-(NO}_2\text{)Tyr}$; $X' = 2\text{-Abz}$

*10 minute couplings: 60°C, DMF, DIC/HOBt
All Fluorophores coupled easily (10 minutes).*

*Dabcyl/FAM Peptide
(>90% purity)*



(a). Peptide Chemistry and Microwave Heating

60°C is optimal (rapid couplings and no by-products)

Single couplings needed in all cases tested to date

DIC/HOBt performed excellently under these conditions

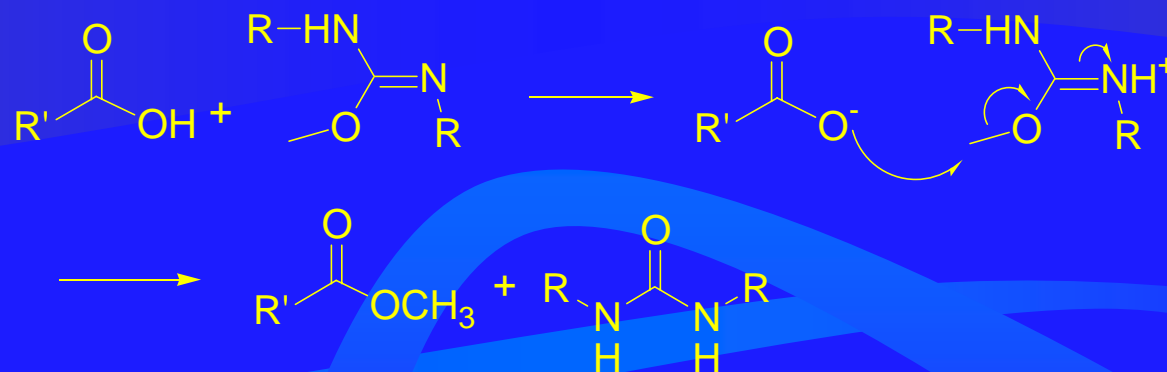
Avoids the use of exotic coupling reagents with their problems and side-reactions

Excellent method for coupling poorly reactive acids such as fluorophores

Time needed reduced from 1 week to 1 day with Peptoid-Peptide conjugate

Arrhenius based acceleration

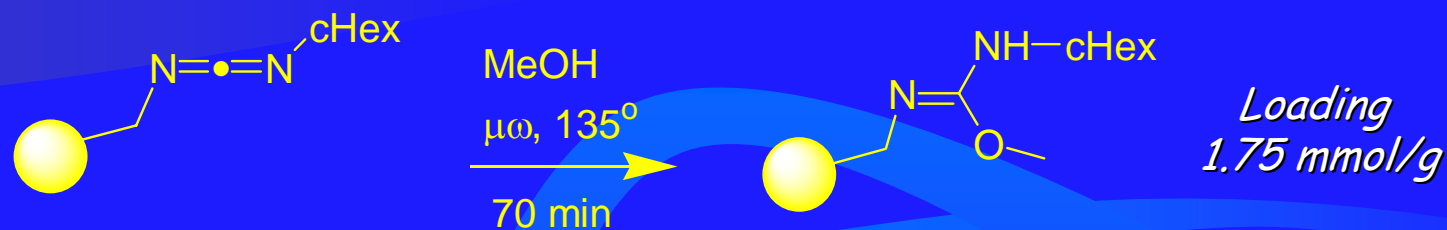
Isoureas as Alkylating Agents (CCE)



Isoureas have been used to prepare esters from carboxylic acids. They are tolerant of other functional groups in the molecule but the urea by-product can be difficult to remove

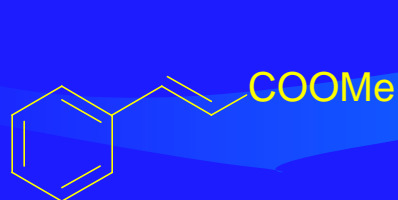
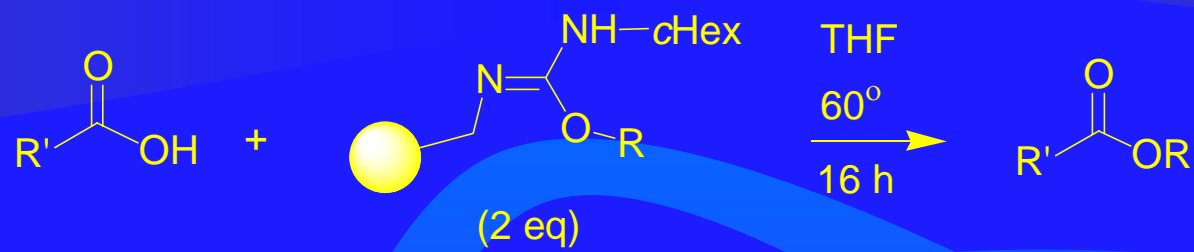
Low toxicity, but also lower reactivity than other alkylating agents

(b). Polymer-Supported O-alkyl-isoureas

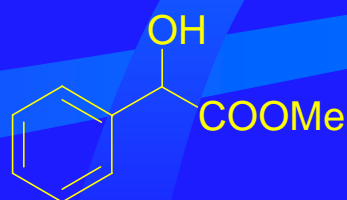


- *Microwave heating with carbodiimide resins*
- *No other reagents apart from methanol are used (no purification problems)*
- *Easily followed by IR*

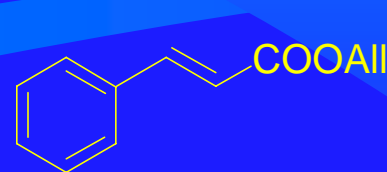
(b). Synthesis of Esters



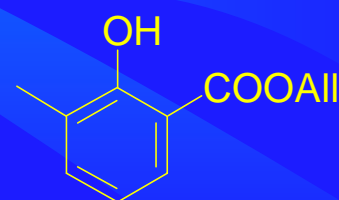
83% (purity >95%)



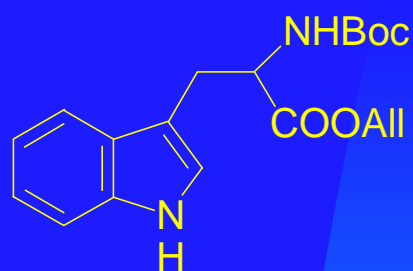
85% (>95%)



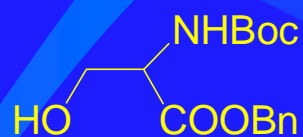
82% (>95%)



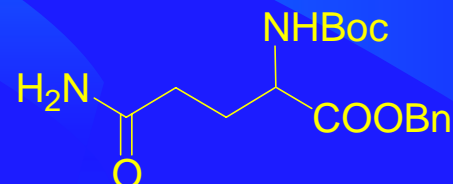
90% (>95%)



93% (>95%)

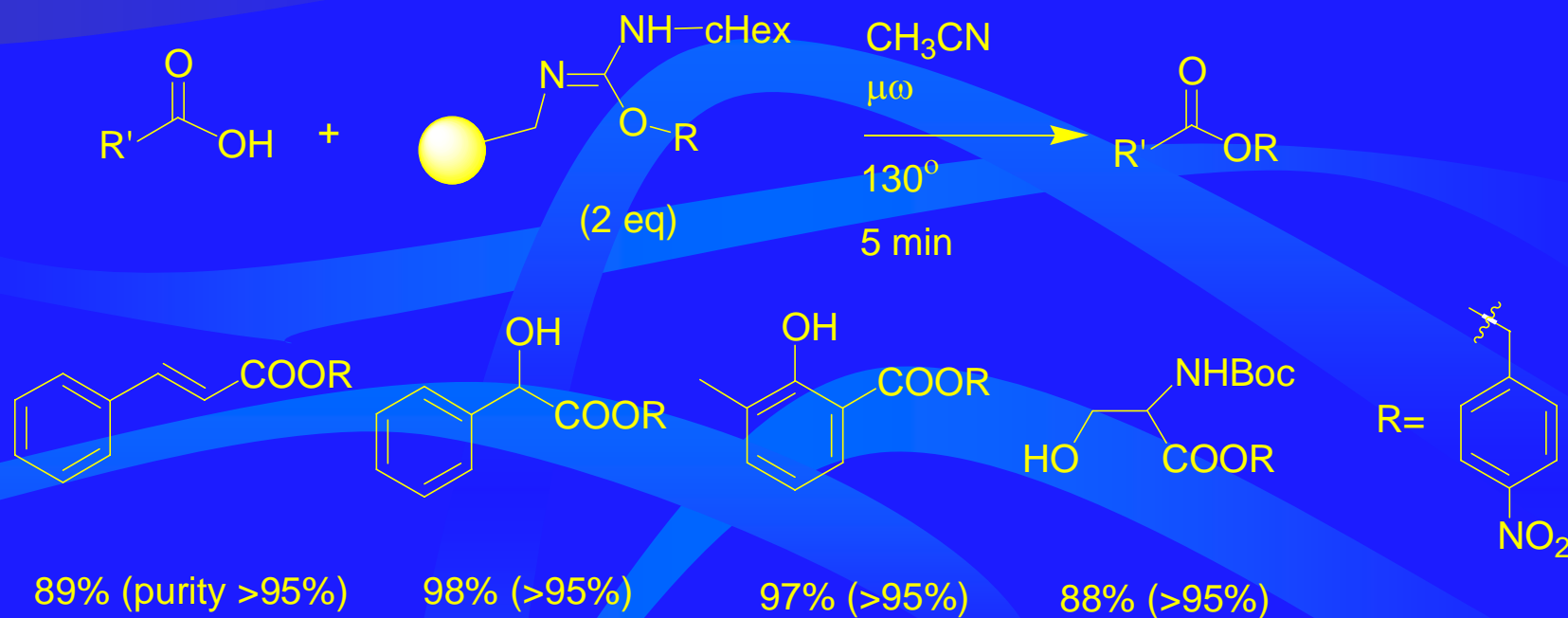


99% (>95%)



98% (>95%)

(b). Microwave-Assisted Synthesis of Esters



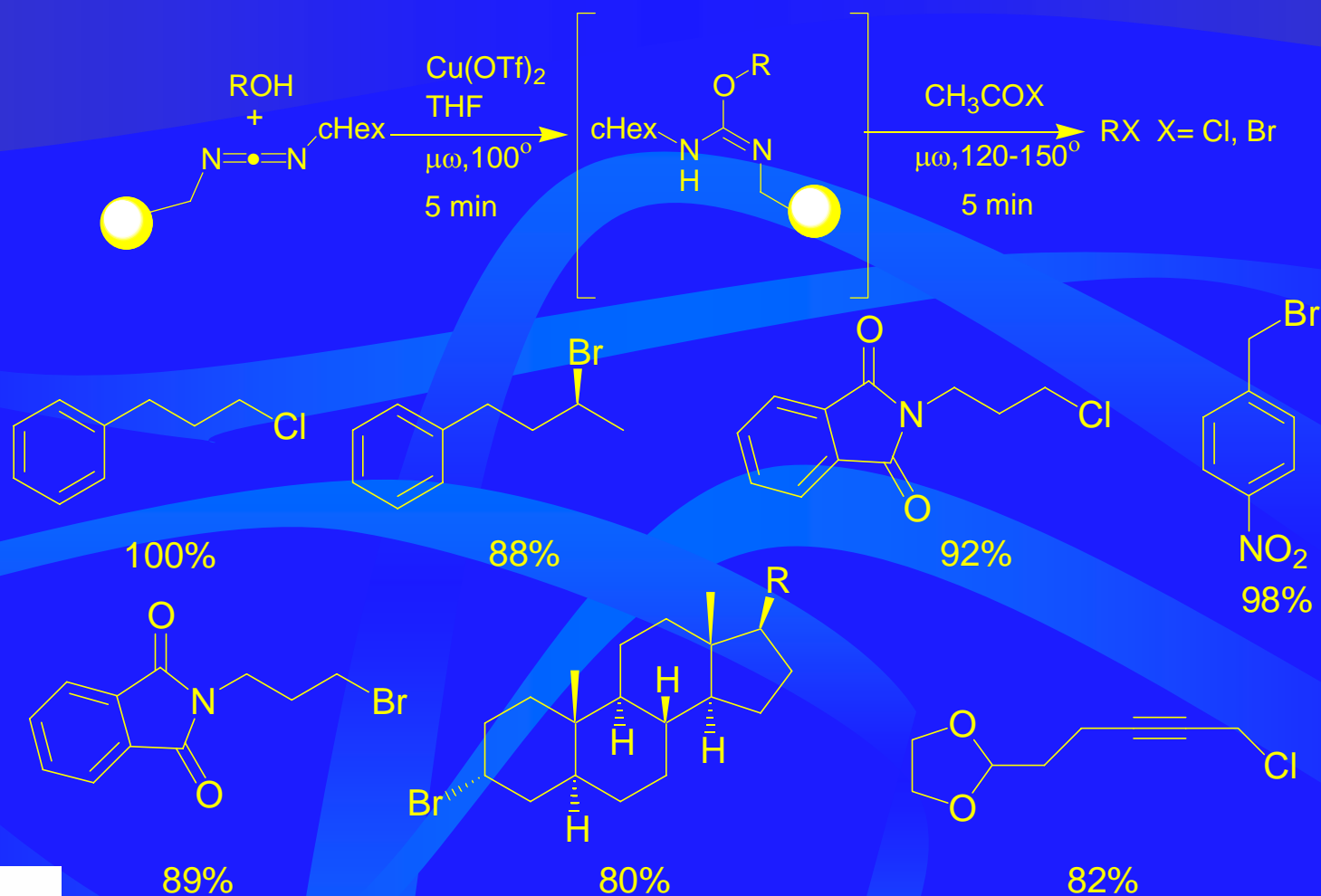
- The use of acetonitrile instead of THF allows faster heating*
- Similar results obtained for methyl, benzyl and allyl esters*

(b). Synthesis of PS-isoureas with Copper (II) catalysis

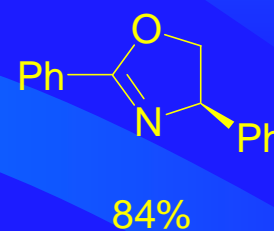
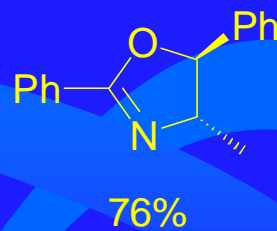
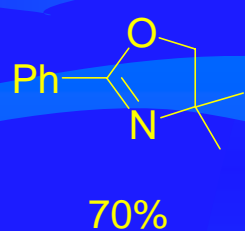
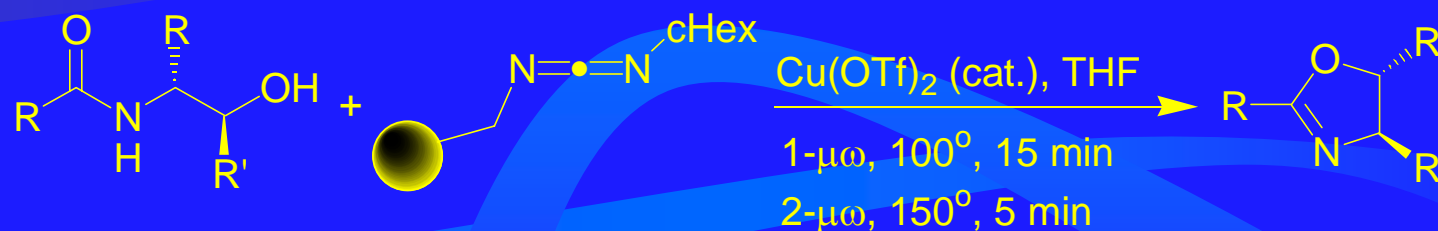


- *The reaction of other alcohols with the carbodiimide can also be performed at RT using $\text{Cu}(\text{OTf})_2$ as catalyst (3-5% w/w)*
- *The copper catalyst removed from the resin using TMEDA.*
- *Washing solution is blue coloured when copper is present, it is easy to detect when the salt is completely removed*

(b). Supported Reagents - PS-isoureas and the synthesis of haloalkanes



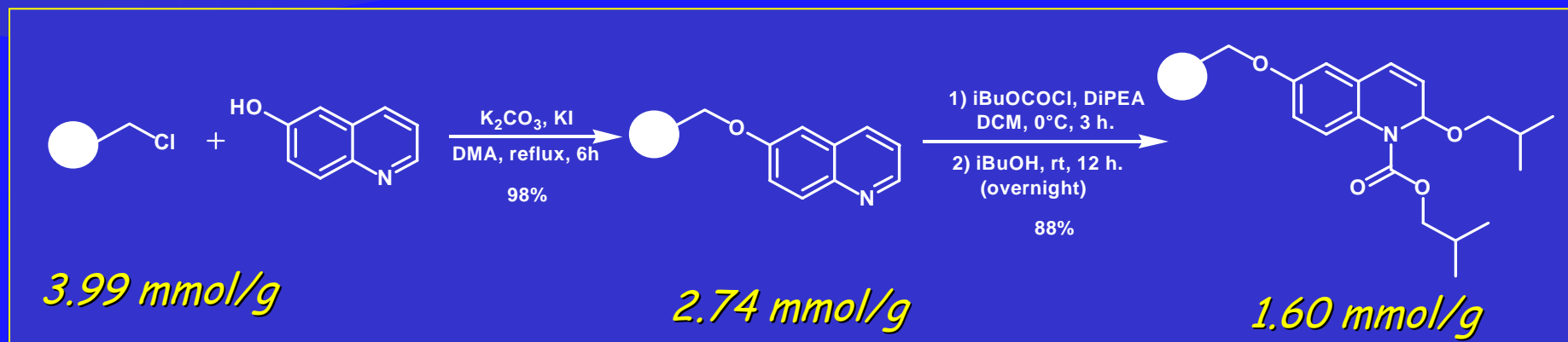
(b). Oxazoline Synthesis via PS-Carbodiimide



- *Products isolated in good yields and excellent purities (>95%)*
- *Filtration on short alumina plug used to remove copper catalyst*

(b). Synthesis of Polymer-Supported IIDQ

Three steps from Merrifield resin :



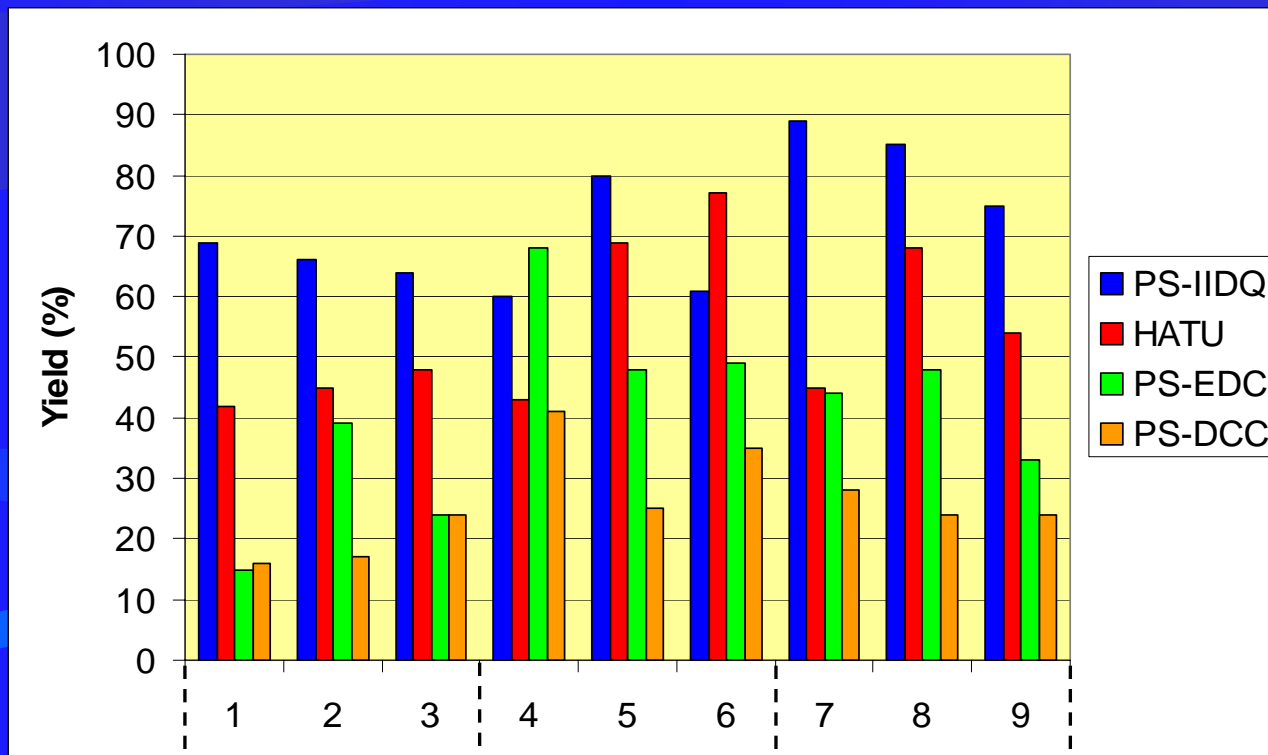
86% overall conversion

No requirement of a pre-activation step

- The order of addition of the amine, acid or coupling agent makes no difference to the efficiency of the coupling reaction*

(b). Polymer-Supported IIDQ: Comparisons

Coupling reagents tested: *PS-IIDQ, HATU, PS-EDC, PS-*



tert-Butylaniline
Benzylamine
PhG-OMe

tert-Butylaniline
Benzylamine
PhG-OMe

tert-Butylaniline
Benzylamine
PhG-OMe

(b). Polymer-Supported IIDQ

Acids	Amines								
	morpholine	cyclohexylamine	Aminoisobutyric acid	4-nitroaniline	tetrahydronaphtylamine	4-tert-Butylaniline	Benzylamine	Phenylglycine	Proline
Aminoisobutyric acid	Yellow	Yellow	Green	Red	Green	Green	Yellow	Yellow	Orange
Phenylacetic acid	Green	Yellow	Green	Red	Green	Yellow	Green	Yellow	Yellow
Benzoic acid	Green	Green	Green	Red	Green	Green	Green	Green	Yellow
Alanine	Green	Green	Green	Red	Yellow	Green	Green	Green	Green
Phenoxy acetic acid	Green	Yellow	Green	Red	Green	Green	Green	Green	Green

Average isolated yield: 73%

■ $Y \geq 67\%$
■ $34\% \leq Y < 67\%$
■ $0 < Y < 34\%$
■ No reaction

(b). Polymer-Supported IIDQ

Characteristics :

PS-IIDQ obtained easily in three steps from Merrifield resin

High loading resin (1.4-1.6 mmol/g)

Stable under normal laboratory storage conditions

Efficiency:

Works well in most cases of general amide bond formation

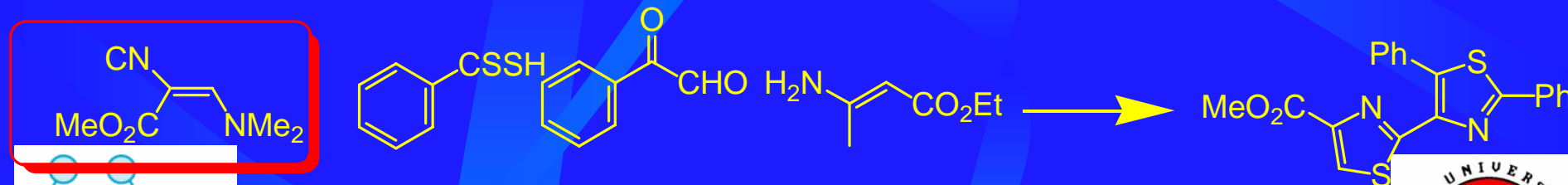
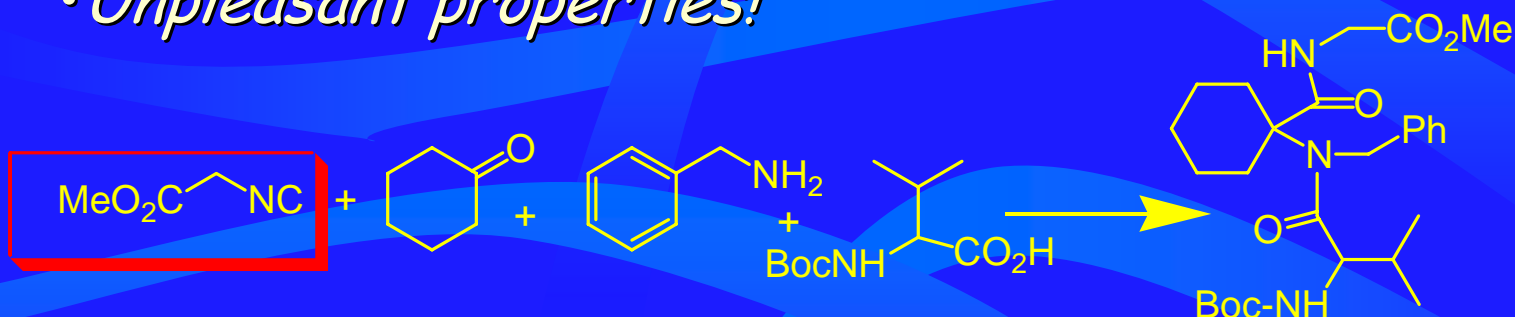
Tests at Evotec OAI on substituted coumarins confirmed the best results over other supported coupling reagents

Limitation :

Hindered amines (Proline type) give carbamate species when coupling to carboxylic acids containing Methyl groups in the α position

(b). Multi-Component Reactions

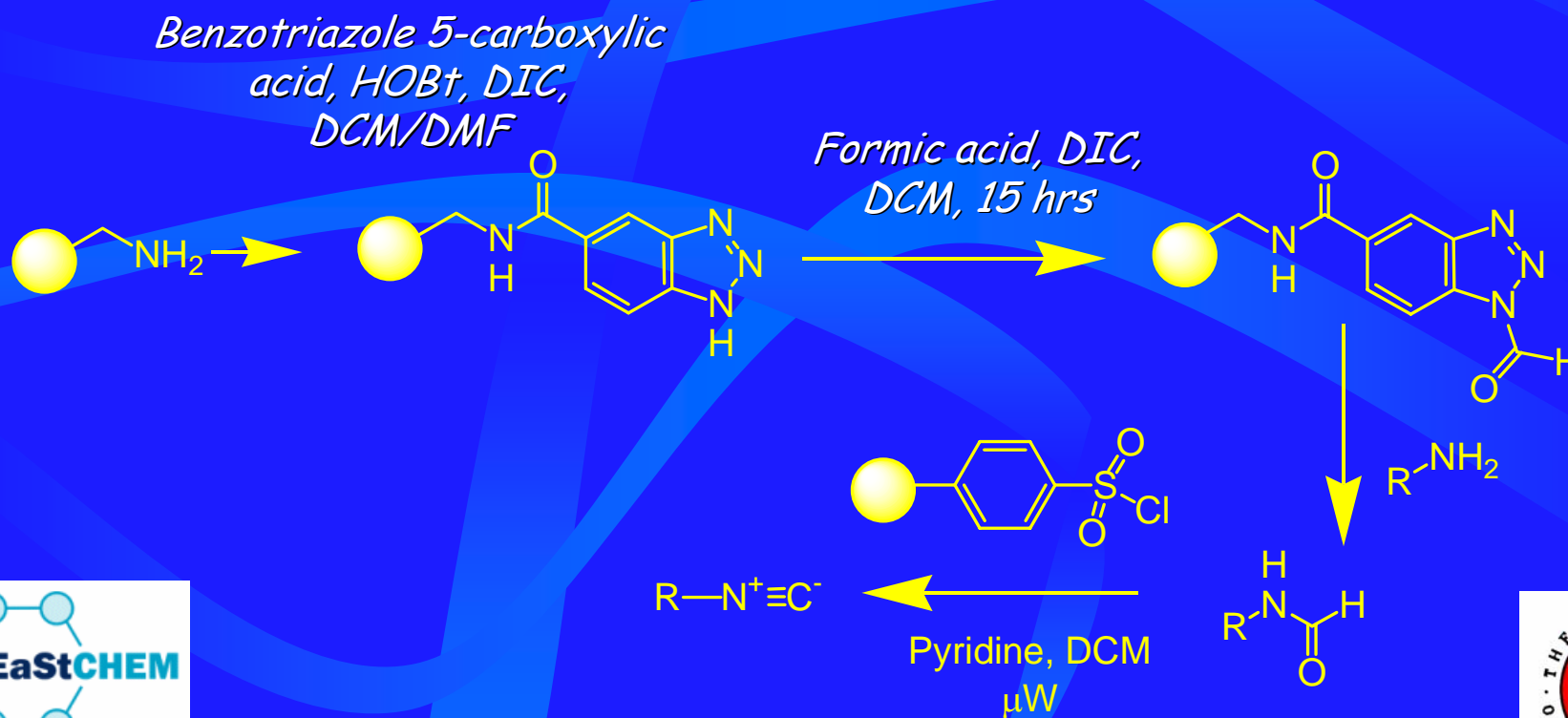
- *Isonitriles are important constituents of many MCR's due to their unusual "split reaction personality"*
- *Limited number commercially available*
- *Unpleasant properties!*



A. Domling - Current Opinion in Chem Biol

(b). Supported Reagents - Isonitrile Synthesis

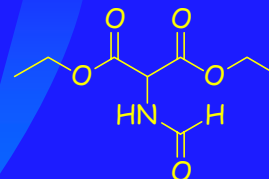
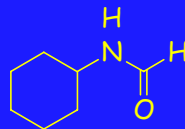
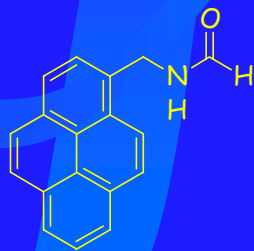
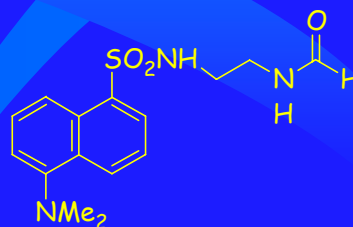
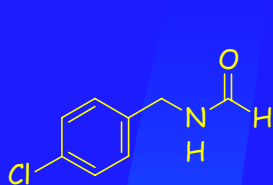
- Isonitriles are important constituents of many MCR's due to their unusual "split reaction personality"*
- Development of new methods for the synthesis of isocyanides using polymer-supported reagents*



(b). Isonitrile Synthesis

Entry	Formamide	Sulfonyl chloride resin	Pyridine in DCM	Isocyanide (% purity)
A	1	3 eq.	50 eq.	100 %
B	1	1.5 eq.	25 eq.	76 %
C	1	2 eq.	50 eq.	90 %
D	2	3 eq.	50 eq.	81 %
E	3	3 eq.	50 eq.	96 %
F	4	3 eq.	50 eq.	100 %
G	5	3 eq.	50 eq.	95 %
H	6	3 eq.	50 eq.	18 %

Microwave: 100°C for 10 minutes.



(c). Heterogeneous Pd Catalysts

- Recover and reuse without loss of activity*
- Increase stability under air and/or moisture*
- Compatible with most of solvents especially water*
- Increase of activity for aryl halide*

Cost & availability: $I < Br < Cl$

Reactivity: $I > Br > Cl$

- Easy to separate and handle*

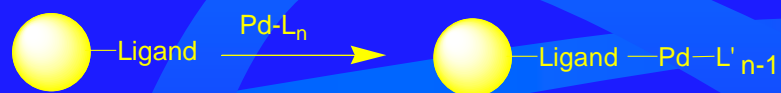
(c). Heterogeneous Pd Catalysts

(1) Conventional method

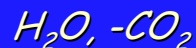
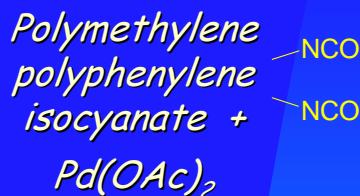
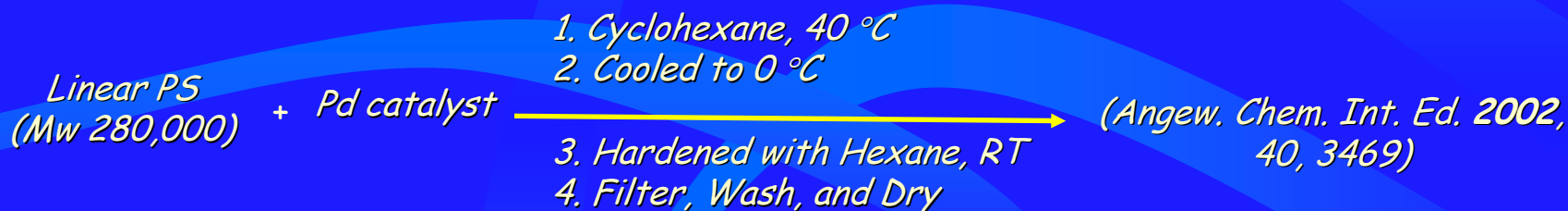
(a) Pd on activated carbon (Pd/C)

(b) Pd adsorption on inorganic supports (silica, alumina, etc)

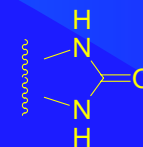
(c) Coordination of Pd catalyst via resin-bound ligands



(2) Microencapsulation



Oil-in-Water emulsion
polymerization



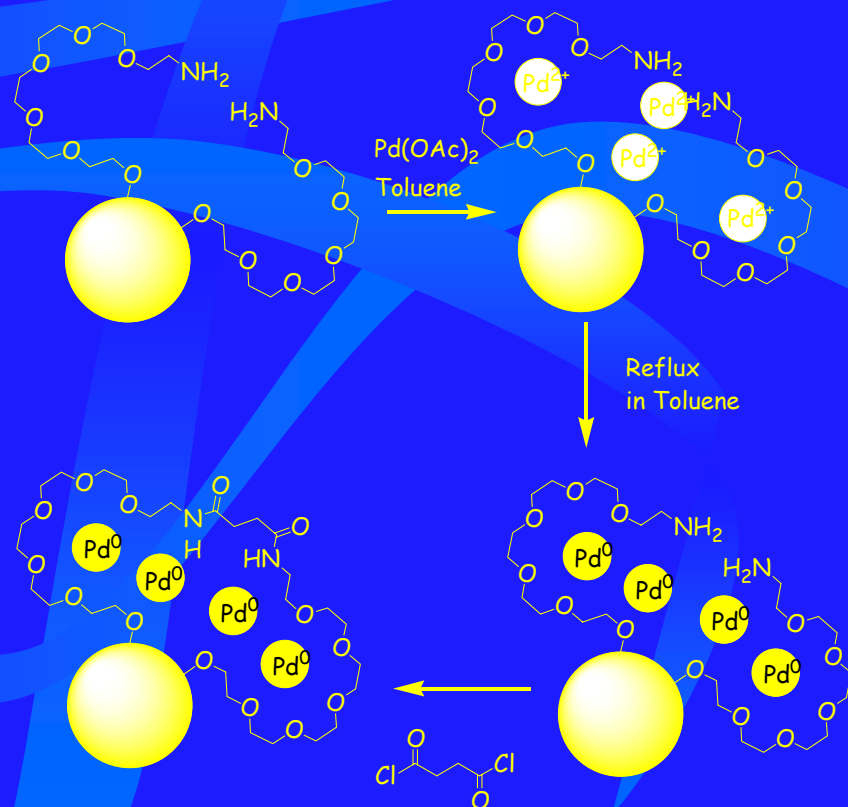
(Chem. Commun. 2002, 1132)

Interaction between vacant orbital of Pd and functionality of polymer backbone e.g. π -orbital of phenyl ring or carbonyl group

(c). Resin-Captured (RC) Pd

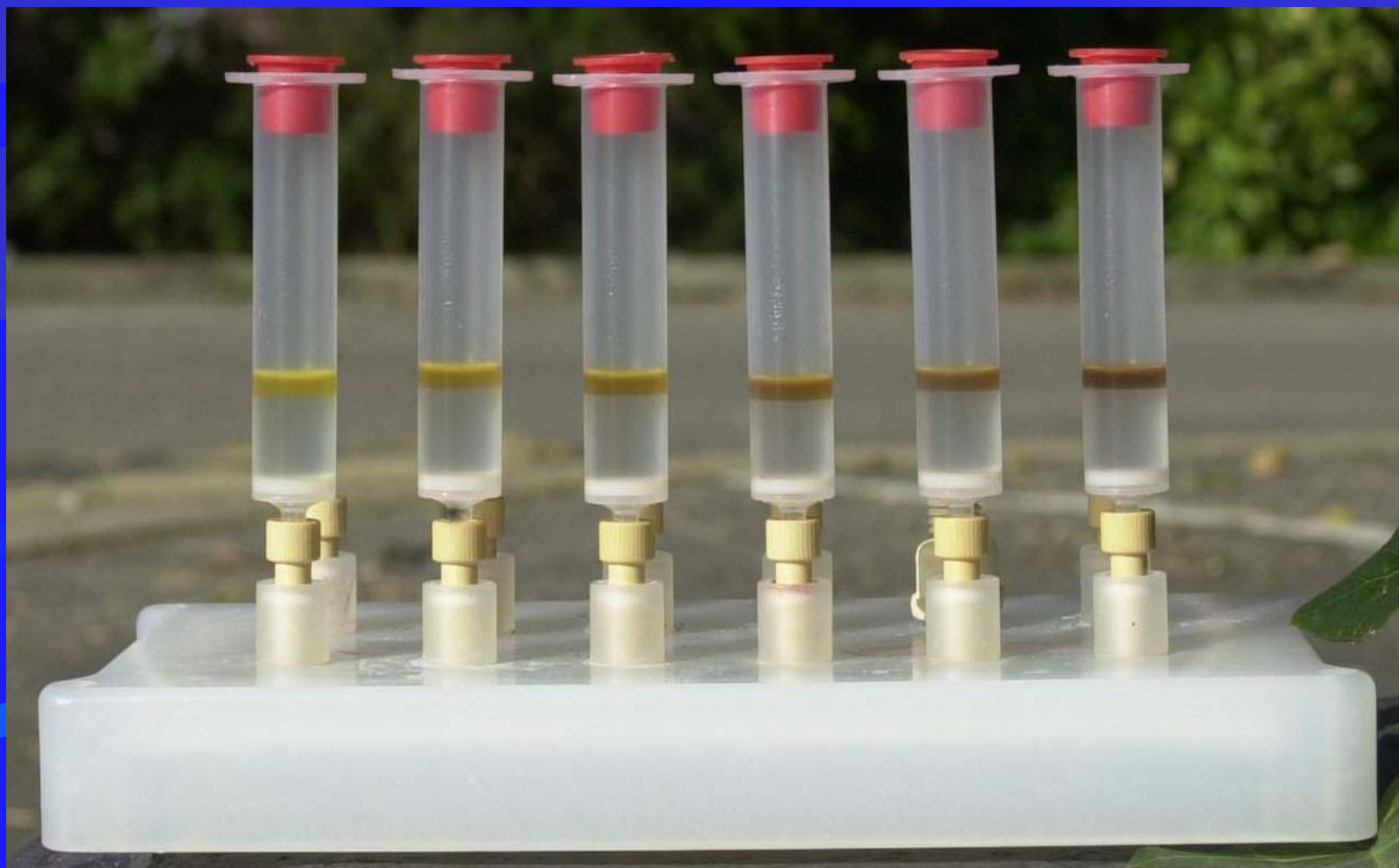
Pd catalyst could be entangled in the form of nano particles on the PS-PEG resin and fixed by cross linking

Interaction between π -electrons of the PS and/or unshared electrons of the PEG and vacant orbitals of Pd



*0.36 mmol of Pd/g
based on Pd analysis
by ICP-AES*

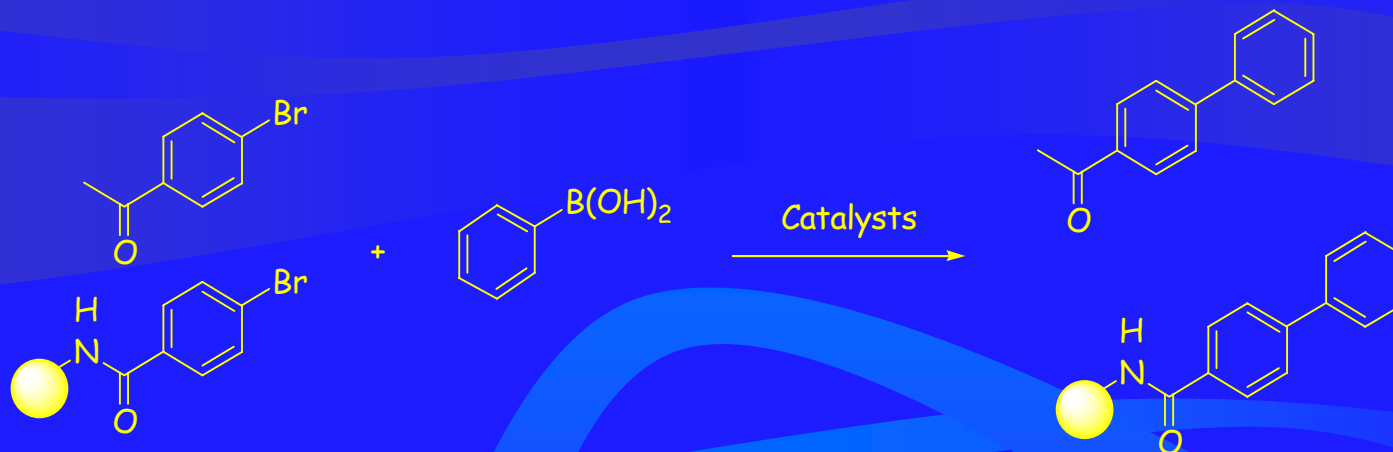
JACS, 2006



Cross-linking %

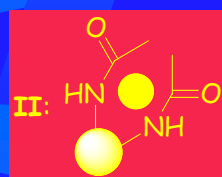
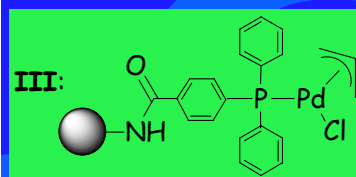
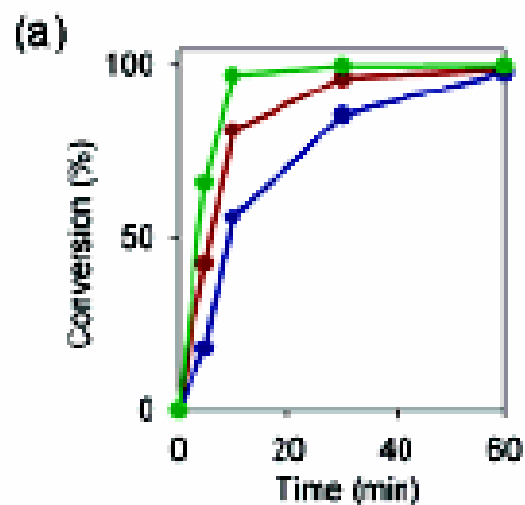
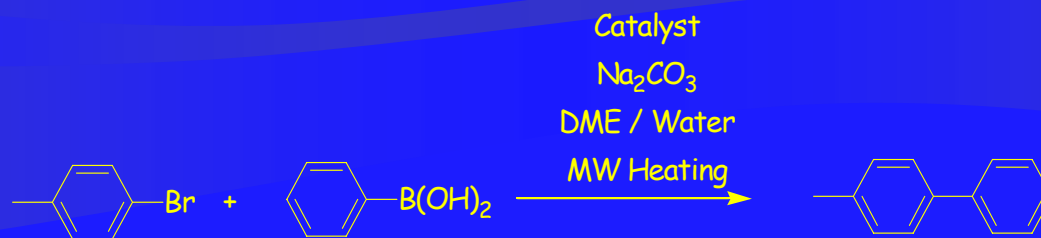


(c). 3 Phase Reaction Test

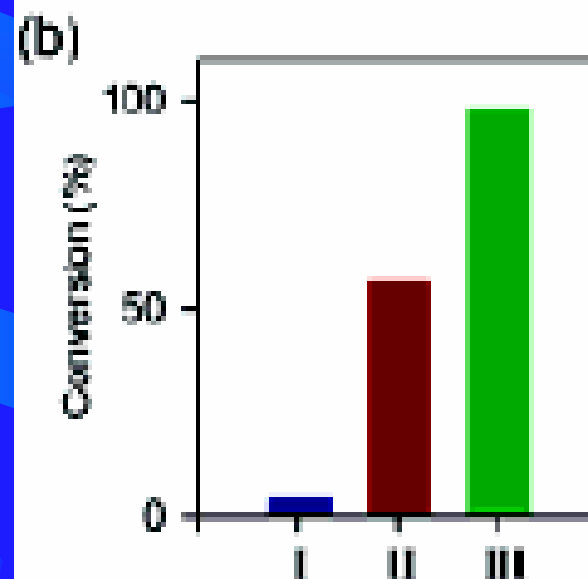


Pd Catalyst	Yield (solution)	Yield (solid)
Pd/C	98%	95%
Pd/Alumina	98%	95%
Pd coordinated by resin-bound phosphine ligand	>99%	85%
Pd captured by resin	99%	17%
Pd captured by resin followed by cross-linking (40%)	98%	4%

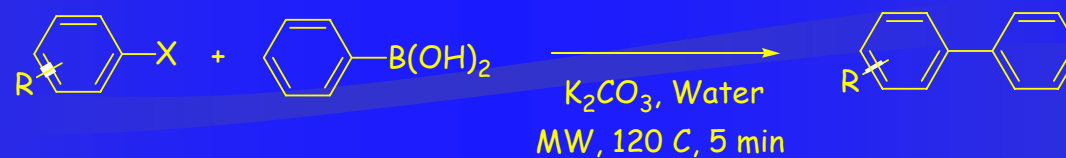
(c). Different Immobilised Pd Catalysts



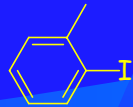
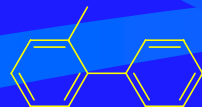
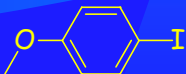

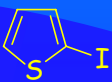
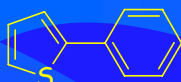



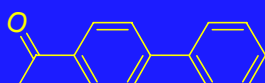
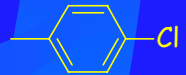



Activity of Filtrate

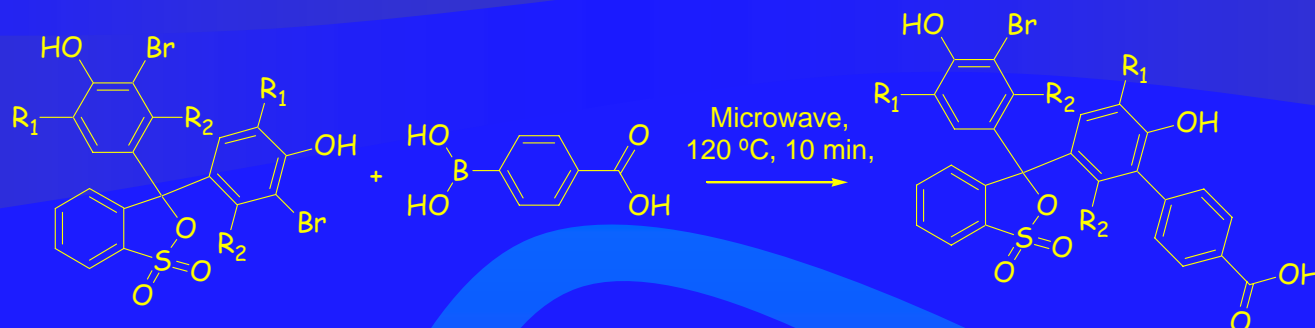


The Pd content in the filtrate from I was 0.09 ppm based on elemental analysis using ICP-OES.



Reactant	Product	Yield (%)
		87
		85
		93
		91
		93
		99
		11

(c). Suzuki Reaction with RC Pd and Crosslinking



	R_1	R_2	yield (%)
Bromophenol blue	Br	H	41 (33)
Bromocresol purple	Me	H	57 (56)
Bromothymol blue	<i>i</i> -Bu	Me	28 (25)

Isolated yield of mono-substituted products and compared with yields obtained using homogeneous catalyst ($Pd(OAc)_2$)

pH Indicator Transitions



Methyl Red



Bromocresol Purple

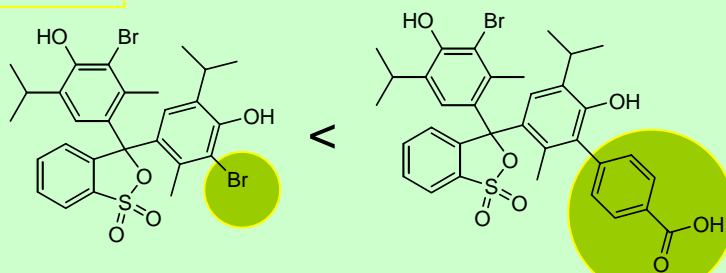


Bromothymol Blue

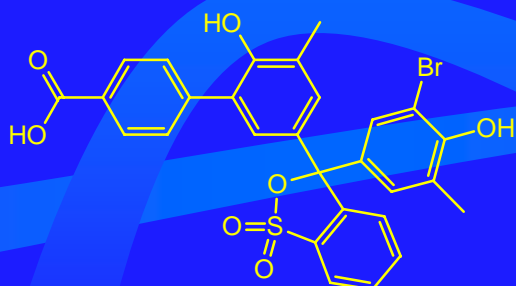
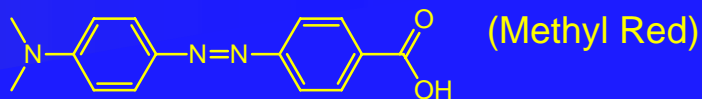
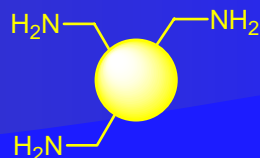


Bromothymol Blue Derivative

pKa



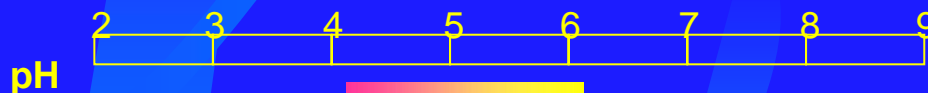
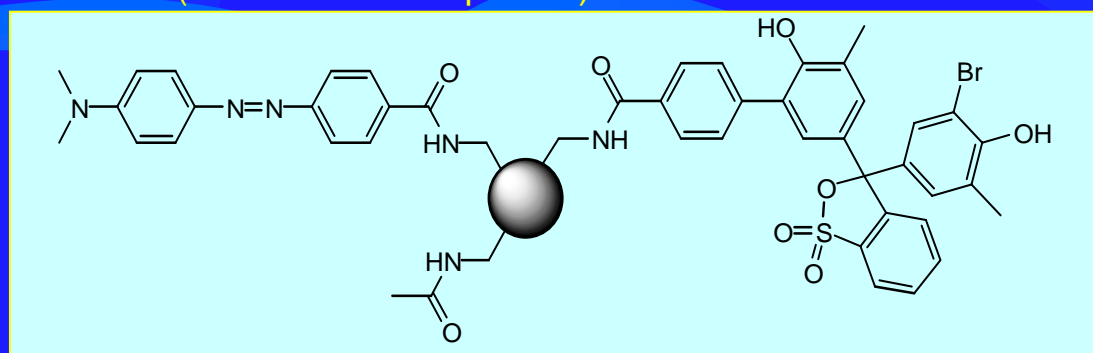
pH-Indicating Resin



(Bromocresol Purple der)

TFFH/TEA used as coupling agents

Capping by acetylation



Methyl Red

Bromocresol Purple

Bromothymol Blue

Bromothymol Blue Derivative

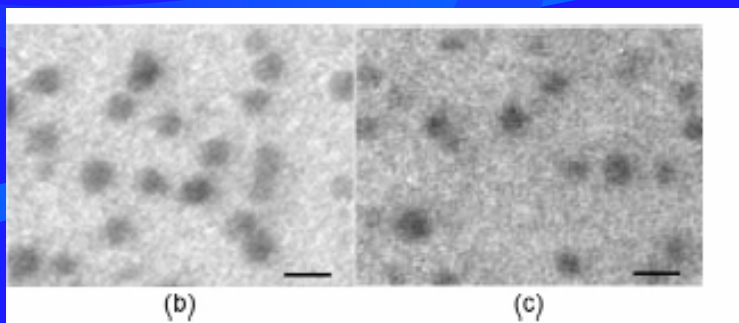
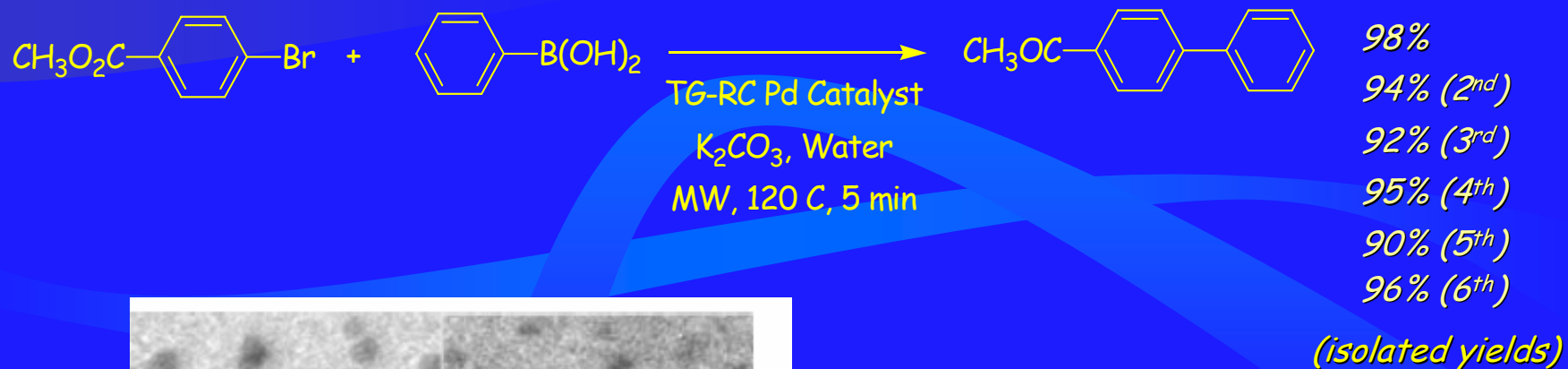
Colour Change of Indicator Resin According to pH



pH

1	3	5	7	9
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(c). Recycling of RC Pd for Suzuki Reactions

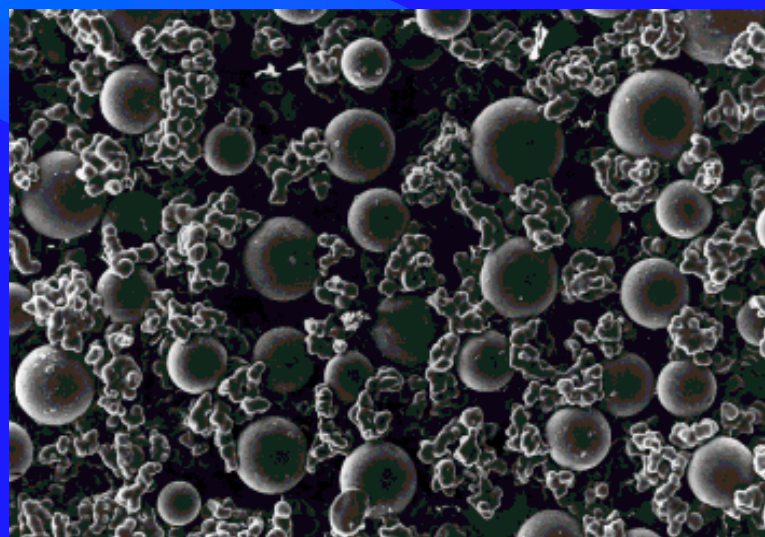
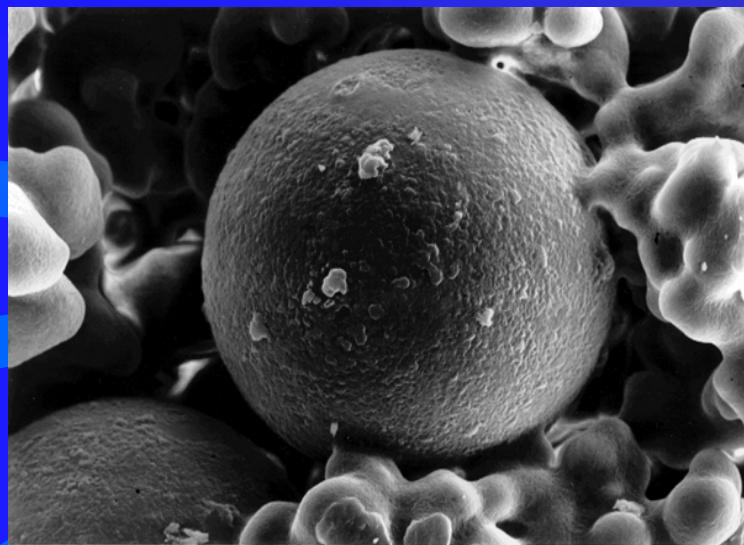
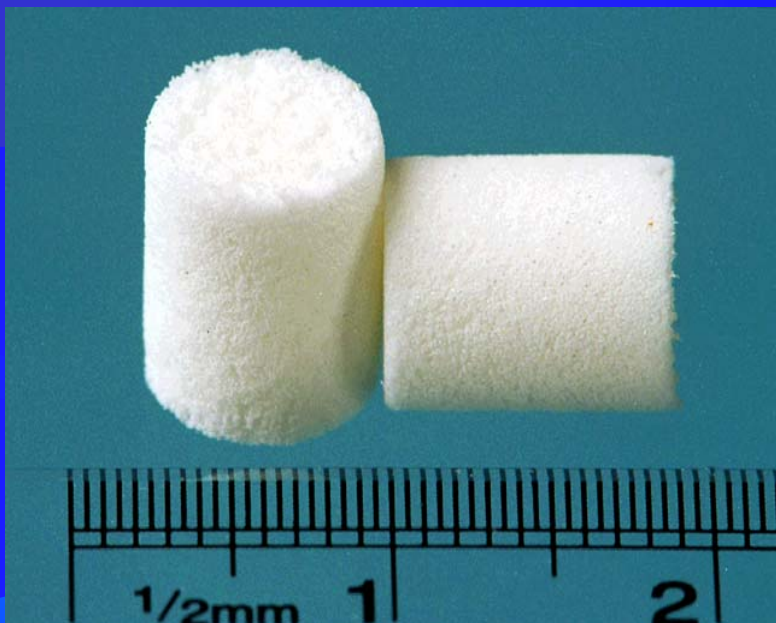


*Pd Nanoparticles
(Left) Before and (Right) After 6
Reuses*

(b). Heterogeneous Pd Catalysts

- Recover and reuse without loss of activity*
- Increased stability under air and/or moisture*
- Compatible with most of solvents including water*
- Easy to separate and handle*

Resin Plugs



*Resin composites
Polystyrene resin and
inert porous filler*

Flow-Through Synthesis

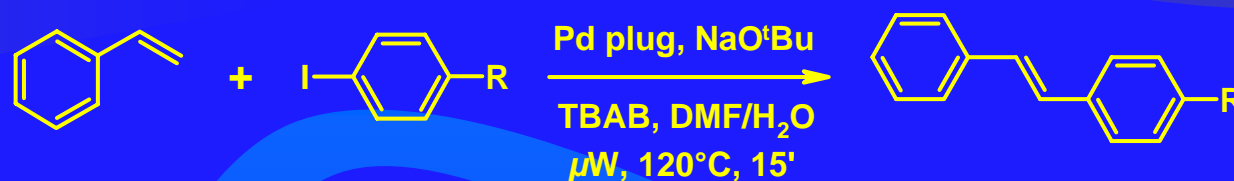


*5-10 cartridges in a single
Column*

*Catalysts, Reagents,
Scavengers etc...*



(c). Plugs - Modular Resin Captured Pd



R	Catalyst quantity (mol %)	Isolated yield (%)
OMe	6.7	93
CF ₃	0.7	94
NO ₂	6.7	79
Me	2.7	96
H	6.7	94
COMe	2.7	82

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