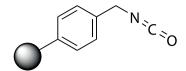
Biotage® MP-Isocyanate

Macroporous Nucleophile Scavenger

(Also see PS-Isocyanate)



Key Facts

















Shelf Life

Capacity (mmol/g)

BSE/TSE

Scalable

Particle Size (µm)

Thermally & Mechanically Stable

Good Laboratory Practice

Bulk Density (q/L)

Specifications

Chemical Name: Polystyrene methyl isocyanate

Resin Type: Highly cross-linked macroporous

poly(styrene-co-divinylbenzene)

Application: Scavenging nucleophiles, including

amines and alkoxides

Scavenging Conditions: 2–3 equivalents relative to

nucleophile, 1-16 h, 20 °C

Compatible Solvents: Non-acidic organic solvents and

vater

Storage: Cool, dry location

MP-Isocyanate is a macroporous polystyrene-bound scavenger, for nucleophiles such as amines and alkoxides. Resin-bound scavengers are added after a reaction is complete in order to quench and react selectively with excess reactants and/or reaction by-products. The resulting resin-bound adducts are removed by simple filtration.

Biotage supplies two isocyanate scavenger resins:
PS-Isocyanate and MP-Isocyanate. The resin backbone in
PS-Isocyanate consists of 1% cross-linked polystyrene-codivinylbenzene. Nucleophiles to be scavenged gain access
to the isocyanate sites by diffusion through the polystyrene
gel. The MP-base copolymer is a highly cross-linked a robust
and low-swelling material, which makes it ideal for restricted
volume environments (e.g. microwave vials and 96-well plates
etc.). Its unique pore structure provides greater access to the
reactive sites without the need for solvent swelling, resulting in

faster reactions and higher recoveries. The abrasion-resistant matrix has better handling characteristics and reduced transfer losses. Both these scavengers can be applied to similar chemistries; the main criteria for differentiation is the intended solvent choice and the format preference.

Representative Procedures

Scavenging Amines using MP-Isocyanate and PS-Isocyanate in 1,2-Dichloroethane

This study compares the ability of PS-Isocyanate and MP-Isocyanate to scavenge four amines from solution in DCE. Three equivalents of scavenger were used relative to the amine.

At ambient temperature, the scavenging rates correlated directly with amine nucleophilicity (Table 1). Both resins were able to scavenge piperidine and benzylamine from solution within 1 h. Scavenging of aniline took place more slowly, reaching only 60–63% completion within 1 h. At 60 °C, PS-Isocyanate scavenged aniline to 92% completion, while MP-Isocyanate removed it completely from solution within 1 h. Scavenging of the highly hindered 2-aminobenzophenone was slow with both scavengers, even at 60 °C.

The Effect of Different Solvents on Scavenging by MP-Isocyanate and PS-Isocyanate

Scavenging in THF, which swells 1% cross-linked polystyrene, was compared with scavenging in acetonitrile and methyl tertbutyl ether (MTBE). Acetonitrile and MTBE are poor solvents for swelling 1% cross-linked polystyrene. MTBE was chosen as a method for other ethereal solvents commonly employed in organic synthesis.



Scavenging rates for both scavengers followed the order THF > Acetonitrile > MeOH > MTBE (Table 2). Benzyl amine was scavenged rapidly in THF reaching 93% and 98% completion within 2 h with PS-Isocyanate and MP-Isocyanate, respectively. MP-Isocyanate scavenged more rapidly and more completely in the three other solvents, demonstrating it to be the scavenger of choice in solvents that do not swell 1% cross linked polystyrene.

Reducing the amount of scavenger from 3.5 equivalents to 2 equivalents reduced the amount of benzyl amine scavenged within the given time. We recommend the use of 3.5 equivalents for complete scavenging, but there is much scope for further optimization, depending on the application.

Electrophile	Temp (°C)	Resin	% Scavenged in 1h	% Scavenged in 3.3 h	% Scavenged in 18 h
Piperidine	RT	PS-Isocyanate	100	100	-
		MP-Isocyanate	100	100	-
Benzylamine	RT	PS-Isocyanate	100	100	-
		MP-Isocyanate	100	100	-
Aniline	RT	PS-Isocyanate	60	91	91
		MP-Isocyanate	63	85	100
Aniline	60 °C	PS-Isocyanate	92	-	100
		MP-Isocyanate	100	-	100
2-Aminobenzophenone	RT	PS-Isocyanate	13	30	83
		MP-Isocyanate	34	57	90
2-Aminobenzophenone	60 °C	PS-Isocyanate	13	-	77
		MP-Isocyanate	39	-	90

Table 1. Comparative scavenging of amines by PS-Isocyanate and MP-Isocyanate in 1,2-dichloroethane.

Solvent	Time (h)	PS-Isocyanate % Scavenged	MP-Isocyanate % Scavenged
THF	2	93	98
	5.5	96	100
	20	97	100
Acetonitrile	1	69	100
	4	91	-
MTBE	1	0	84
	4	0	100
	20	0	-
МеОН	2	0	66
	5.5	8	75
	20	20	78

Table 2. Scavenging benzylamine by 3.5 equivalents of PS-Isocyanate or MP-Isocyanate in various solvents at ambient temperature in different solvents at room temperature. 2.0 equivalent of resin was used in these reactions.



Summary

In comparison with PS-Isocyanate, MP-Isocyanate offers the following benefits:

- Scavenges effectively in solvents that do not swell 1% cross-linked polystyrene
- » Limited swelling, so can be used where volume is restricted (e.g. 96 well plates, microwave vials)
- » Faster scavenging than PS-Isocyanate
- » For certain substrates MP-Isocyanate offers more complete scavenging than PS-Isocyanate

Ordering Information

Part Number	Quantity
801504	3 g
801409	10 g
801410	25 g
801411	100 g
801412	1000 g

References

- 1. Booth, R. J., Hodges, J. C. J. Am. Chem. Soc. 1997, 119, 4882.
- 2. Gooding, O., Labadie, J.; Porco, J. *J. Comb. Chem.* 1999, 1, 420–422.



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